

## pK<sub>a</sub> Values of Common Bases

Values in H<sub>2</sub>O as much as possible, so common comparisons (i.e., H<sub>2</sub>O pK<sub>a</sub> = 15.7) can still be used

Note; The pK<sub>a</sub> values associated with bases is normally meant to refer to the true pK<sub>a</sub>'s of their conjugate acids; i.e., pK<sub>a</sub> associated with HO<sup>-</sup> is 15.7, which is the pK<sub>a</sub> of H<sub>2</sub>O. This is often sloppily used by organic chemists, and can lead to confusion, especially with amines.

