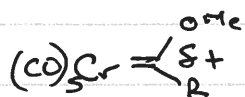


CARBENES / ALKYLIDENES

$L_n M = CR_2$ - METAL C FORMAL DOUBLE BOND.

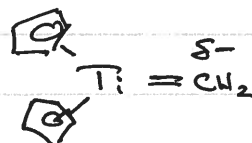
- HISTORICALLY, CARBENES WERE OFFICIALLY CONSIDERED TO FALL INTO ONE OF TWO TYPES - THERE ARE MORE NOW

FISCHER TYPE CARBENES



- MOSTLY Cr GROUP (Cr, Mo, W)
OR Fe GROUP.
- STABILIZED BY ATTACHED HERFROATION
(O HERE)
- REACTIVITY - ELECTRON POOR AT C
- ANALOGY TO ESTER CARBONYLS HAS
BEEN MADE

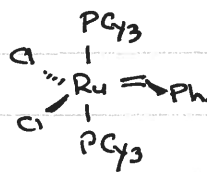
SCHROCK TYPE CARBENES (ALKYLIDENES)



- EARLY TRANSITION METAL
- ELECTRON RICH AT CARBON
PENTAS
- EXAMPLE - TEBBE⁺ REAGENT
(WITTIG R₂N ANALOGUE)

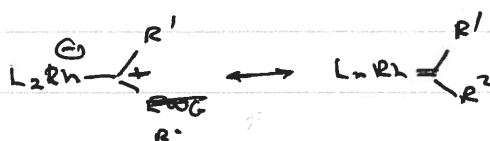
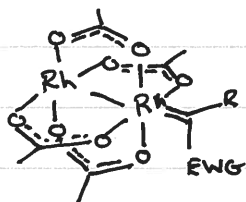
NEWER METATHESIS CARBENES

- SOME Mo^{IV}, MOST Ru^{II} I.C



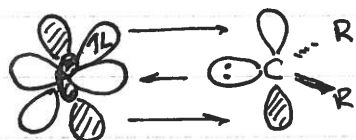
- HAVE SEEN THEM LUMPED IN WITH BOTH TYPES ABOVE
- NO GREAT M-C BOND POLARITY; LIKES ALKENES + CYCLOADDITION PROCESSES.

DIAZO-DERIVED CARBENES - USUALLY Rh^{II} $R-\text{N}_2-R'$ + Rh₂(OAc)₄



- GOOD FOR CYCLOPROPANATIONS, C-H INSERTIONS
- C IS ELECTROPHILIC, DUE TO EWG INvariably
ON CARBENE CARBON.

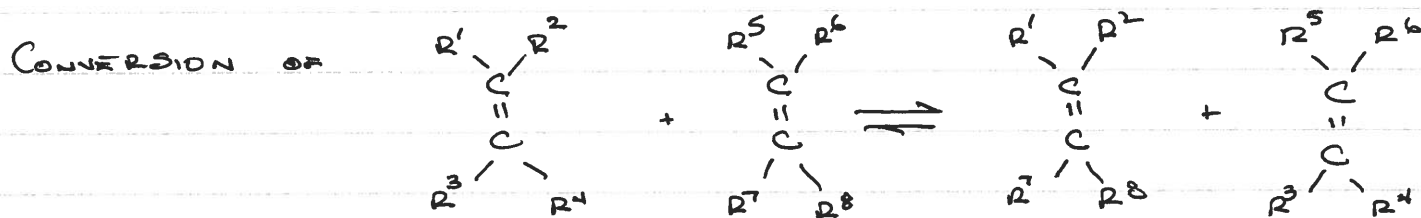
SIMPLEST PICTURE OF BONDING



SINGLET CARBENE

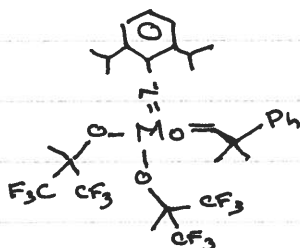
METATHESIS OF ALKENES (+) - BY FAR THE MOST SYNTHETICALLY IMP.

CHAUVIN SCHROCK GRUBBS NOBEL 2005



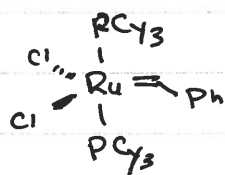
- BY A SERIES OF 2+2 CYCLOADDITIONS AND RETRO- 2+2 CYCLOADDS.
MAY BE CONSIDERED "INSERTION RXNS."

- THE MAIN CATALYSTS



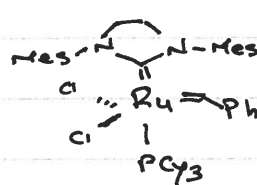
SCHROCK (PRE) CATALYST

- HIGH REACTIVITY
- LESS STABLE, LESS EASILY HANDLED
- NOT V. FUNCTIONAL GROUP TOLERANT



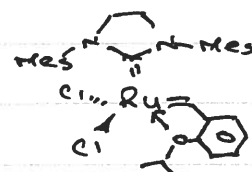
GRUBBS 1ST GENERATION (G1)

- MORE EASILY HANDLED
- V. FUNCTIONAL GROUP TOLERANT
- MUCH LESS REACTIVE THAN SCHROCK



GRUBBS 2ND GEN. (G2)

MORE REACTIVE THAN G1



HOVEYDA-GRUBBS 2 (HG2)

- SOMEWHAT LESS REACTIVE THAN G2
- GOOD FOR E- POOR ALKENES

V. ROUGH REACTIVITY (level)

WILL RIP THE DIENE RIGHT OUT OF YOUR HAND

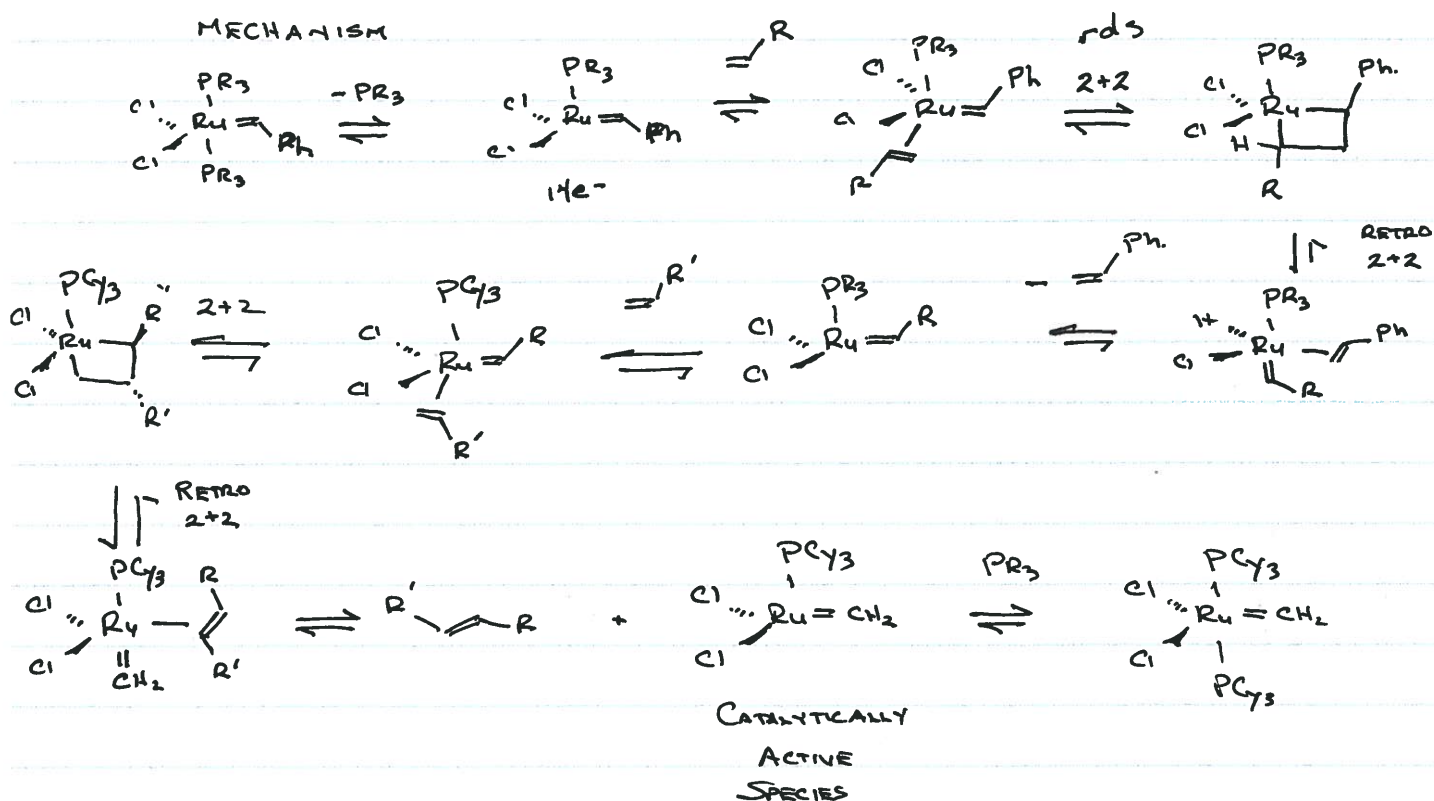
1

65

SOMEWHAT LESS THAN G2

NUMBERS ARE VARIABLE

MECHANISM



- RXN OFTEN DRIVEN BY LOSS AT END OF VOLATILE ALKENE I.E. $\text{H}_2\text{C}=\text{CH}_2$

MAJOR TYPES OF ALKENE METATHESIS

RCM - RING CLOSING METATHESIS

CM - CROSS METATHESIS

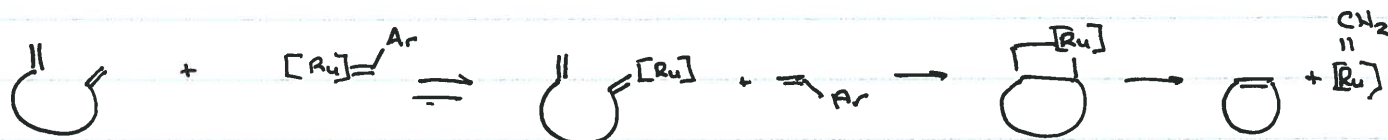
ROMP - RING OPENING METATHESIS POLYMERIZATION

ADMET - ACYCLIC DIENE METATHESIS

WILL INCLUDE BRIEFLY
ENE METATHESIS

WILL NOT DISCUSS
Z. WANG?

RCM - MOST IMPORTANT OF THEM ALL



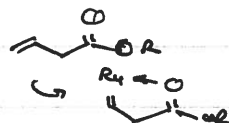
Ru CATALYSTS NORMALLY TOLERATE

ALCOHOLS*, ETHERS*, KETONES, ACIDS, ESTERS*, AMIDES, ARENES

* - ALCOHOLS - ALLYLIC ALCOHOLS CAN BE PROBLEMS IN SLOWER CASES $\rightarrow$ ETHER, ESTER, ACETALS OR

* - VINYL/ENOL ETHERS - ^{OFTEN} TOO UNREACTIVE WITH Ru CATALYSTS

* - ESTERS - OCCASIONAL ISSUES WITH INTERNAL CHELATE BEING TOO STABLE



- ~~3°~~ AMINES



USUALLY MAKE AMIDE OR PG'D AMINE

THIOETHERS



MO OR

PHOSPHINES



MO OR

PROBLEMATIC

FOR NORMAL (5,6) SIZED RINGS.



CATALYST

G1

G2

SCHROCK

DISUB (R=H)

OK

OK

OK

GRUBBS ORG LETT 1999 1, 953

TRISUB (1 R, 2 CH₃)

USUALLY

OK

OK

TETRASUB (2 R'S = CH₃)

USUALLY NOT

USUALLY

USUALLY

J.ORG. CHEM 1997, 62, 7310.

SIZE OF RING.

- 3 & 4 - NO, BECAUSE THERMODYNAMIC FORCE IS TO RING OPEN

- HAS BEEN USED IN ROMP?

- 5, 6 -

- USUALLY SEAMLESS

7 -

FAIR

8 - 10 -

TOUGHEST DUE TO ENTROPY, SEVERAL EXAMPLES, BUT TEND TO USE EVERY TRICK (LOW CONC.; ^{GEM} DIMETHYL/THORPE INGOLD EFFECT!)

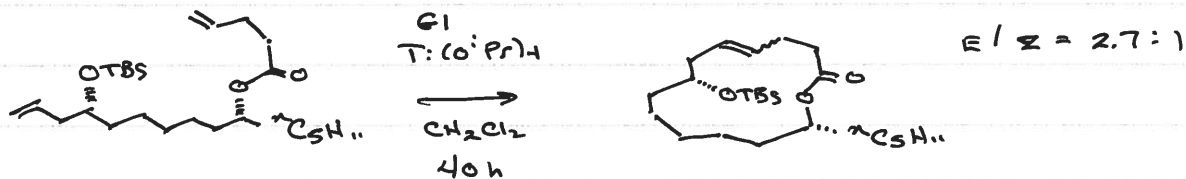
> 10

- NOT TOO BAD - RXNS TYPICALLY AT LOW CONC. (10⁻³M) AND

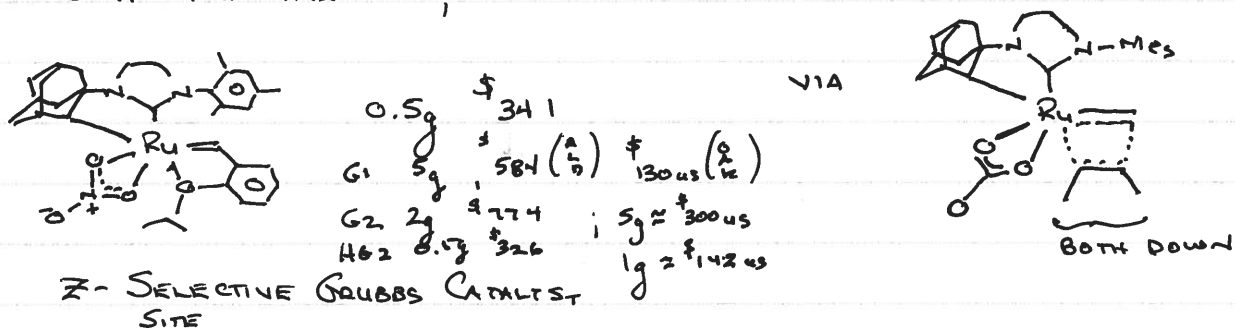
WAIT FOR TRANSIENT OLIGOMERS TO UNRAVEL.

- NOTE: IN RING SIZES ≥ 10 Z-STEREOCHEMISTRY IS NO LONGER MANDATORY, AND Z/E STEREOCHEMICAL CONTROL IS POOR.

i.e.



- THERE HAS BEEN WORK DONE ON CONTROLLING ALKENE STEREOCHEMISTRY, AND IF YOU HAVE \$\$,



- NOTE - ALKENE ISOMERIZATION CAN OCCASIONALLY INTERVENE - BENZOQUINONE ADDED AS A SOLUTION

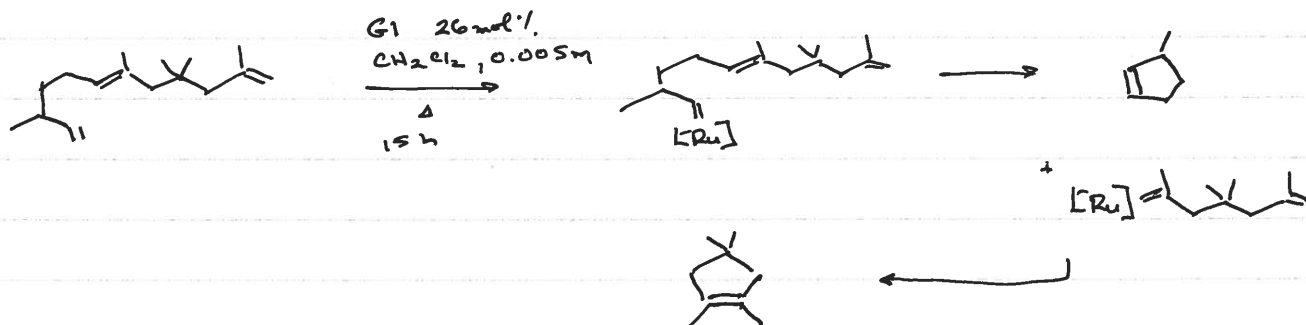
RELAY RING CLOSING METATHESIS

- PROBLEM - IN MANY CASES, THERE CAN BE A DESIRED RCM PRODUCT WHOSE STARTING ALKENES ARE TOO HINDERED TO GO WELL.

i.e.

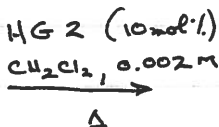


- A SOLUTION TO THIS IS TO CREATE A DIFFERENT STARTING MATERIAL, WITH AN 'ACCESSIBLE' ALKENE, THAT CAN GIVE THE ALKYLIDENE NEEDED FOR RCM INDIRECTLY - CALLED RRCM SOMETIMES



HOYE, J.

J. AM. CHEM. SOC. 2004, 126, 10210.

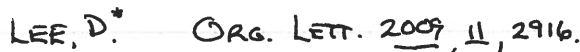

$$R' \approx 4$$

12.

157

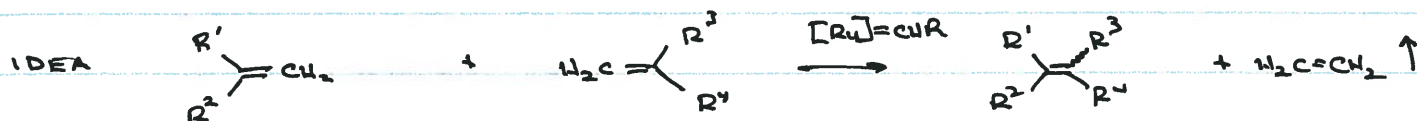
71.6.

PORCO, J.A.
ANGREW. CHEM. INT. ED.
2004, 43, 3601.



CROSS METATHESIS (CM)

- OTHER MAJOR ORGANIC CHEMISTRY USE OF METATHESIS CHEM.



- THE PROBLEM HERE, IS TO GET SELF/HOMO-METATHESIS



- IT IS POSSIBLE TO SIMPLY JUST ADD AN EXCESS OF ALKENE, AND COUNT ON STATISTICS, AND ASSUMING IT IS CHEAP AND THAT ITS HOMO-DIMER IS READILY REMOVABLE

- MORE COMMONLY DONE SELECTIVELY, SUCH THAT WE HAVE A MORE REACTIVE ALKENE (TO METATHESIS), AND A LESS REACTIVE ONE, AND (MOST OFTEN) TO ADD AN EXCESS AMOUNT OF THE LESS REACTIVE ONE

NOTE: IN THIS CHEMISTRY GRUBBS II IS MUCH MORE COMMONLY USED THAN GRUBBS I - HG II USED SOMETIMES W/ e⁻ POOR ALKENES

- GRUBBS HAS CLASSIFIED ALKENES AS TYPE I THROUGH TYPE IV

TYPE I - ALKENES THAT DO FAST HOMODIMERIZATION; DIMERS WILL RE-ENTER CATALYTIC CYCLE

TYPE II - ALKENES THAT HOMODIMERIZE SLOWLY; DIMERS WILL REENTER CATALYTIC CYCLE ONLY RELUCTANTLY

TYPE III - ALKENES THAT DO NOT HOMODIMERIZE, BUT WILL ENTER METATHESIS W/ MORE REACTIVE ALKENES

TYPE IV - ALKENES COMPLETELY UNREACTIVE TO ANY METATHESIS, BUT CAN WATCH

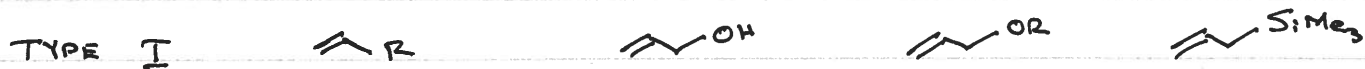
- CROSS METATHESIS IS USUALLY PERFORMED WITH A TYPE I ALKENE AND AN EXCESS OF A TYPE II OR A TYPE III ALKENE

- I WOULD NOT RULE OUT EXAMPLES OF A TYPE II WITH XS TYPE III

GIVE TABLE FROM GRUBBS, R.H. J. AM. CHEM. SOC. 2003, 125, 11360.

- A SHORTENED VERSION

- FOR GRUBBS 2



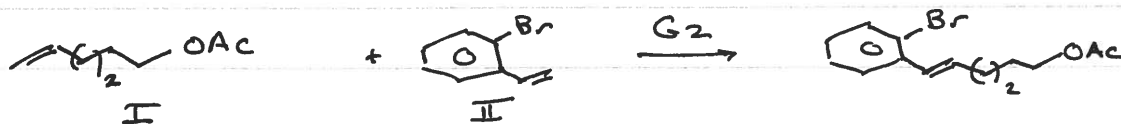
R = Allyl, Ac



- USUALLY GET "MOSTLY" E, BUT E/Z MAY NOT BE GREAT

- OF COURSE, FOR G1 AND HG2 THE TYPE I - TYPE IV LIST WILL BE DIFFERENT

HG2 "ALWAYS" USED WITH $\text{CH}_2=\text{CH-CN}$ ACRYLONITRILE, AND GET A SUBSTANTIAL AMOUNT OF Z ISOMER



AMOUNT? 1 : 1 80% UNUSUALLY GOOD
 1 : 3 "98%"