

Progress and Challenges in Using LIBS for Bacteriological Identification

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PRESENTS

SCIX2015

THE GREAT
SCIENTIFIC EXCHANGE

Sept. 27-Oct. 02 | Rhode Island Convention Center Providence, RI

*LIBS FOR PHARMACEUTICAL AND BIOLOGICAL
DIAGNOSTICS*

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University of Windsor

Outline

- Review of Current State of the Art
- Update on Our Group's Methodology
- New Results / New Questions?
- Concluding Thoughts

Progress on Using LIBS for Bacterial Identification

2007-
2012

The bacterial LIBS spectrum for a given species is stable and **does not change with time** (experiments conducted on the same *E. coli* strain over the course of multiple years).

2007&
2012

Bacterial identification appears to be **independent of the growth condition** and culture medium in which the bacteria were grown.

2011

This result confirmed (*Marcos-Martinez et al. Universidad Complutense, Madrid*) on three similar growth media

2011

Salmonella enterica serovar Typhimurium identified at various concentrations in various liquids such as milk, chicken broth, and brain heart infusion. (*Barnett et al. Alabama State*)

Progress on Using LIBS for Bacterial Identification

2011

Bacterial LIBS spectra do not change with time as the bacteria age/sit on an abiotic surface

2013

This result confirmed (*Multari et al. ARA*) on cutting board, sink drainer

2012

Bacteria can be identified with high sensitivity and specificity when specimens are **obtained from clinical samples** (e.g. sterile urine containing organic and inorganic solutes) without the need to remove other compounds present in the sample.

2012

Live pathogenic *Bacillus anthracis* Sterne strain and *Francisella tularensis* can be **differentiated regardless of mounting protocol** (as lawn and/or colonies on agar, dilutions on agar, and dilutions on glass slides.) (*Multari et al. ARA*)

Progress on Using LIBS for Bacterial Identification

2011
2012

Bacteria in **mixed samples are identifiable**. The dominant or majority bacterial component of a two-component bacterial mixture is reliably identified provided it comprises 70% of the mixture or more. Trace mixture or contamination is insignificant.

2011

Bacterial LIBS spectra can be obtained from **killed** (via autoclaving) or **inactivated** (via UV light) **specimens**, and such treatment (which renders the specimen completely safe for handling) **does not decrease identification specificity** and does not decrease LIBS spectral intensity.

2013

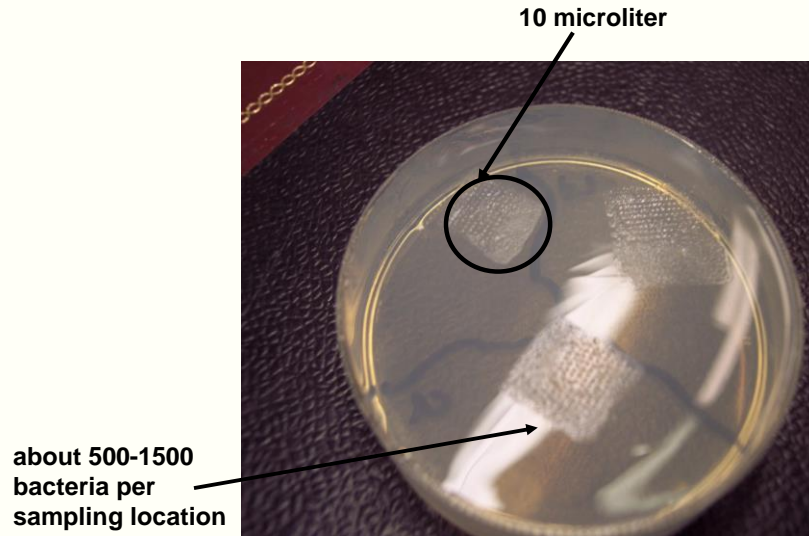
Heat killed bacteria are differentiable (*Multari et al., ARA*)

2015

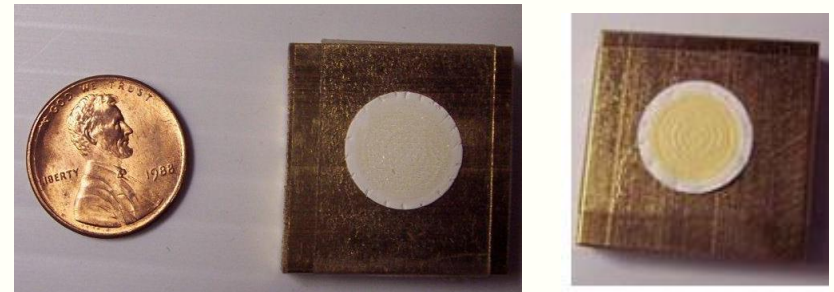
Inactivation by sonication / autoclaving is differentiable (*Sivakumar et al., Delaware State*) with fs-LIBS & ns-LIBS. Changes in spectral intensity were observed.

New Bacteria Testing Procedure

Previously: mounted on agar



Currently: mounting on nitrocellulose bacteriological filter

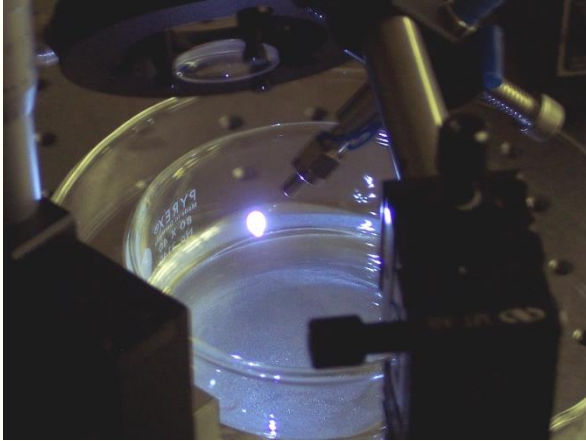


✓ *Advantages: Flatter. Easier to do. More reproducible. Less time.*

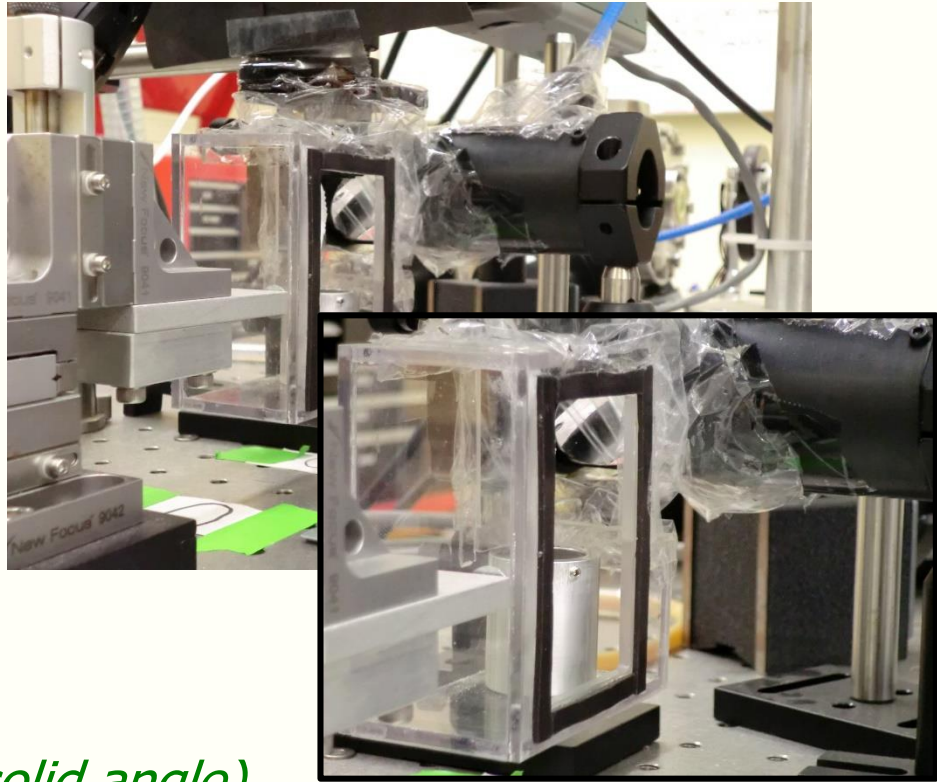
x *Disadvantages: Carbon background.*

New Bacteria Testing Procedure

Previously: light collected by fiber



Currently: using matched parabolic reflectors into fiber



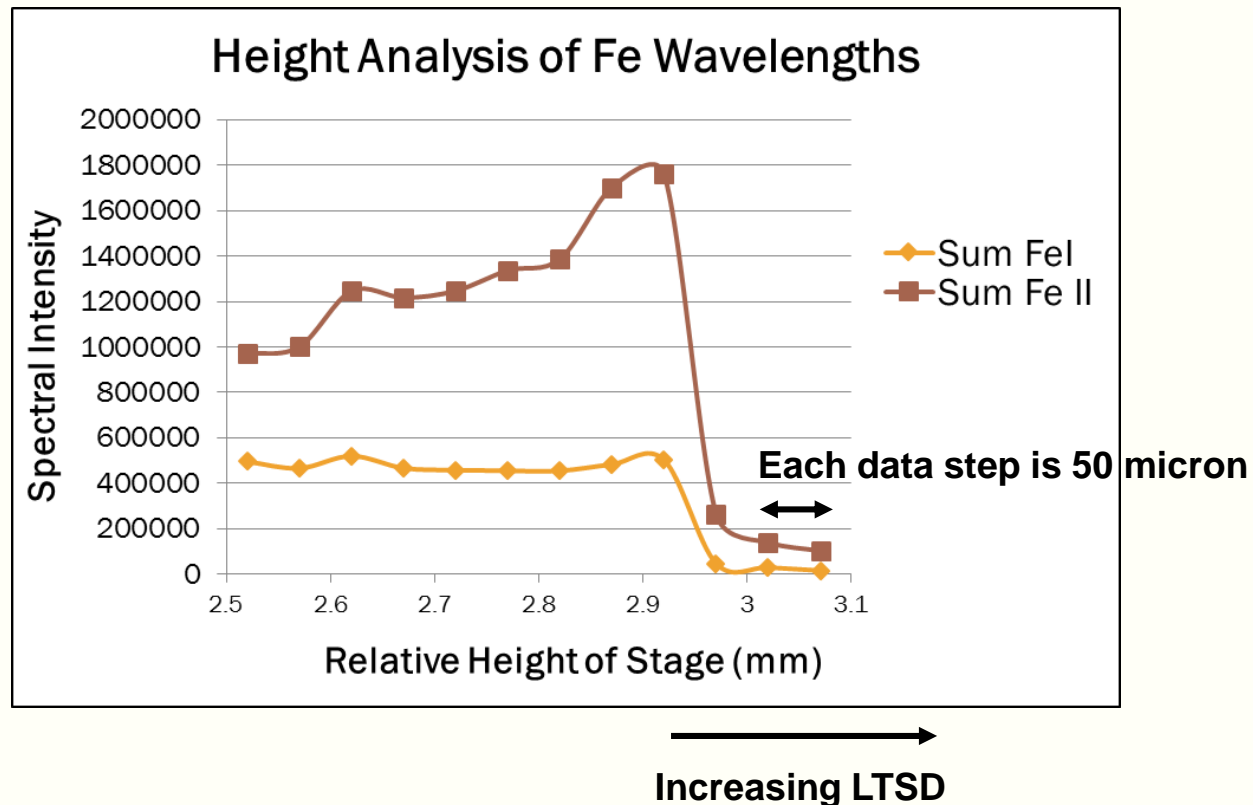
✓ *Advantages: More light collected (solid angle)*

x *Disadvantages: More sensitive to vertical position*

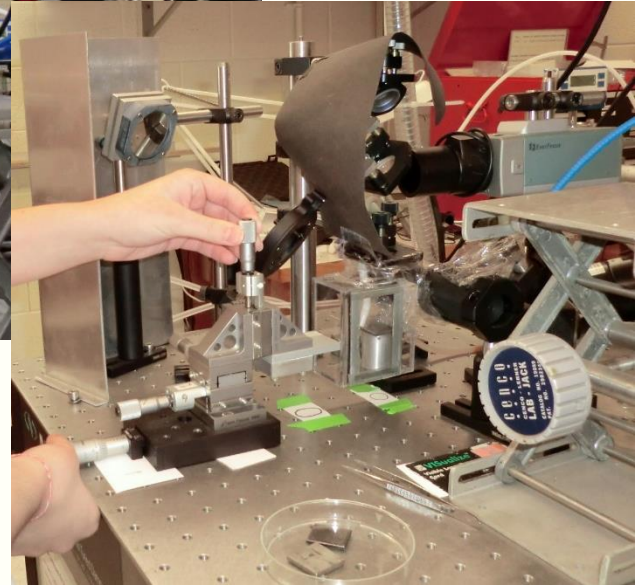
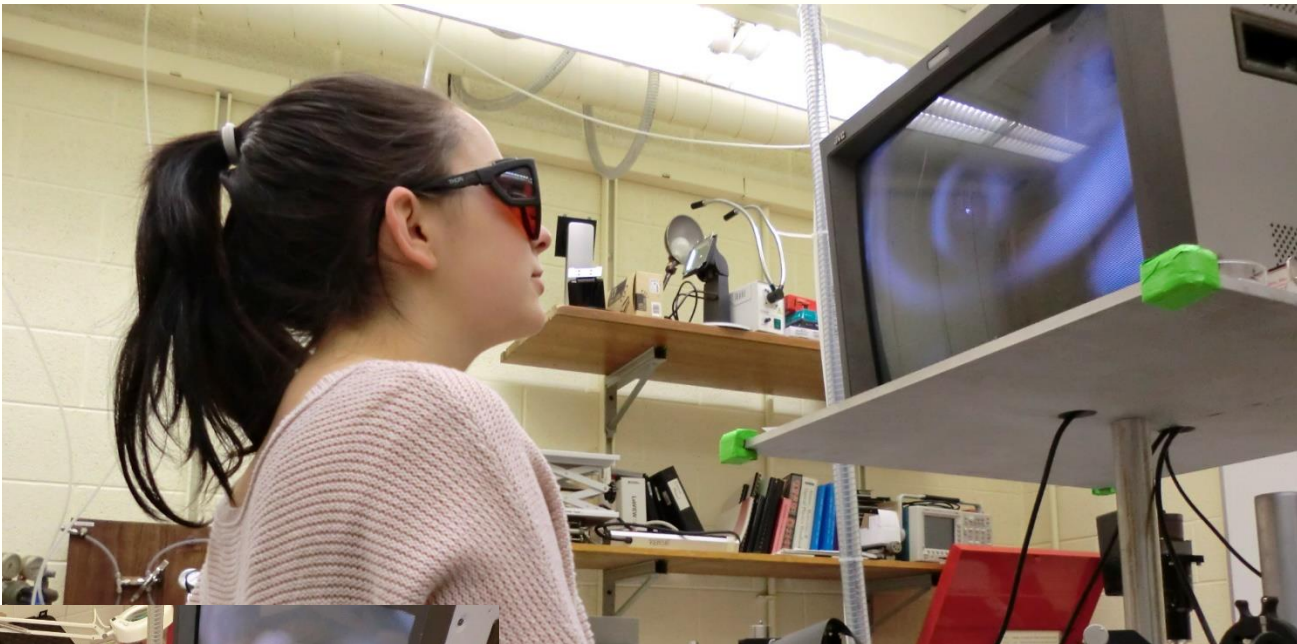
New Bacteria Testing Procedure

Previously: proper LTSD found by trial and error

Currently: Appropriate LTSD/focus found with laser indicator.
Calibrated steel sample tested every day for intensity & spot size.
Unnormalized intensities very reproducible and controlled.



Student-Actuated Focus Finder



Current Method

Our Method of Bacteria Classification

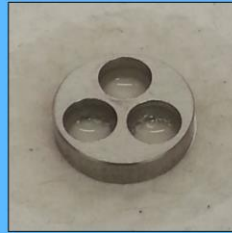
Bacteria is cultured using trypticase soy agar (TSA).



Colonies are removed and placed in 1.5 mL distilled water.



30 μ L of vortexed sample are deposited on a standard 0.22 μ m cellulose filter in contained wells.



Colloidal solution is dried forming a bacteria lawn on the clinician-friendly filter.



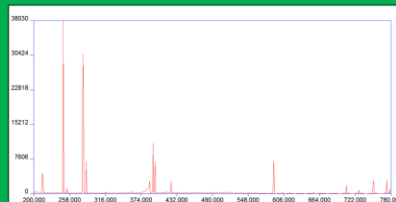
Filter is placed in an argon environment and ablated using a pulsed 1064 nm Nd: YAG laser.



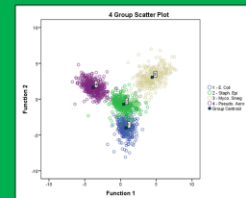
Average time to complete bacterial classification = 1 hour



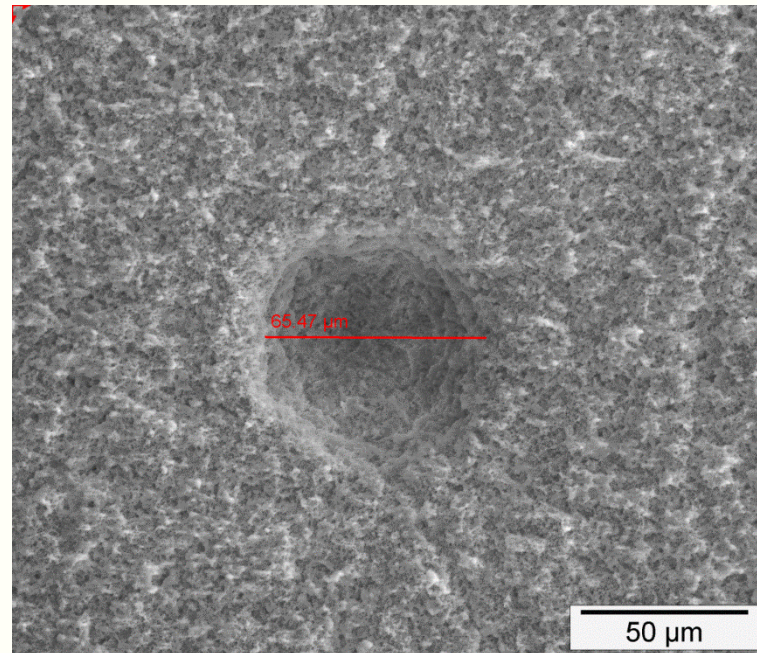
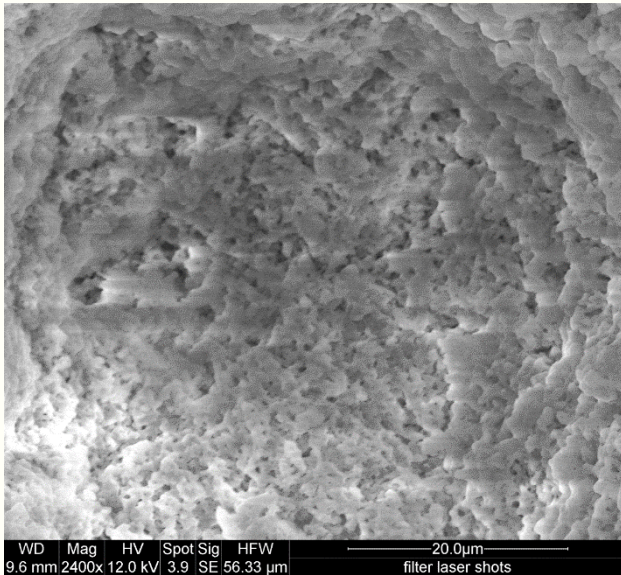
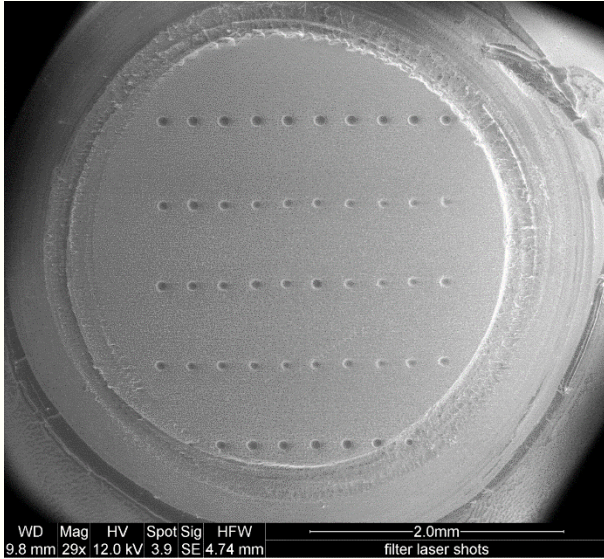
Échelle diffraction grating spectrometer is used to obtain the atomic spectrum and composition of sample.



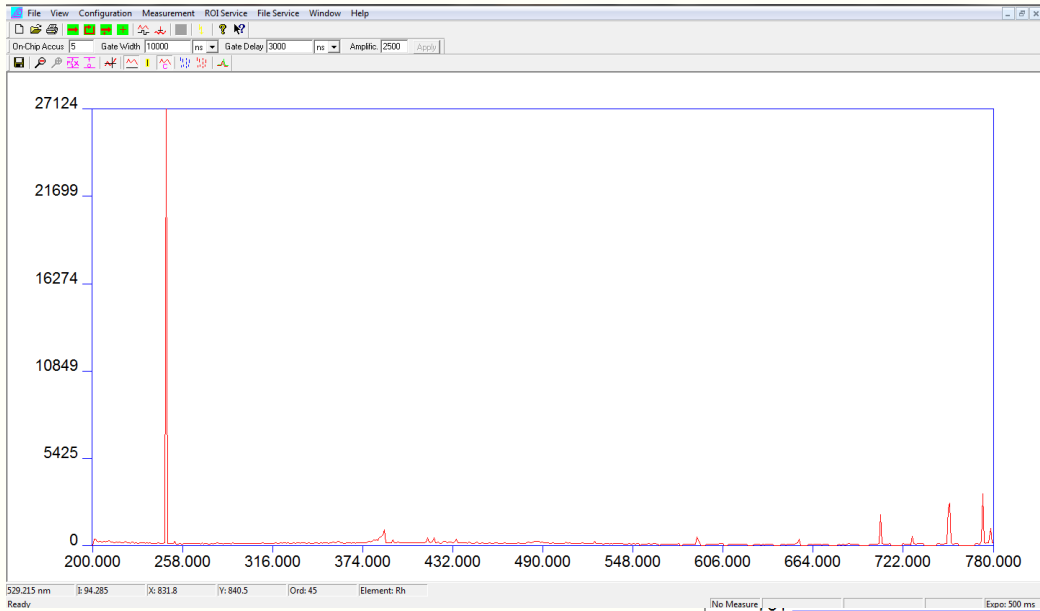
Atomic composition is used to discriminate bacteria against pre-existing library.



Filter

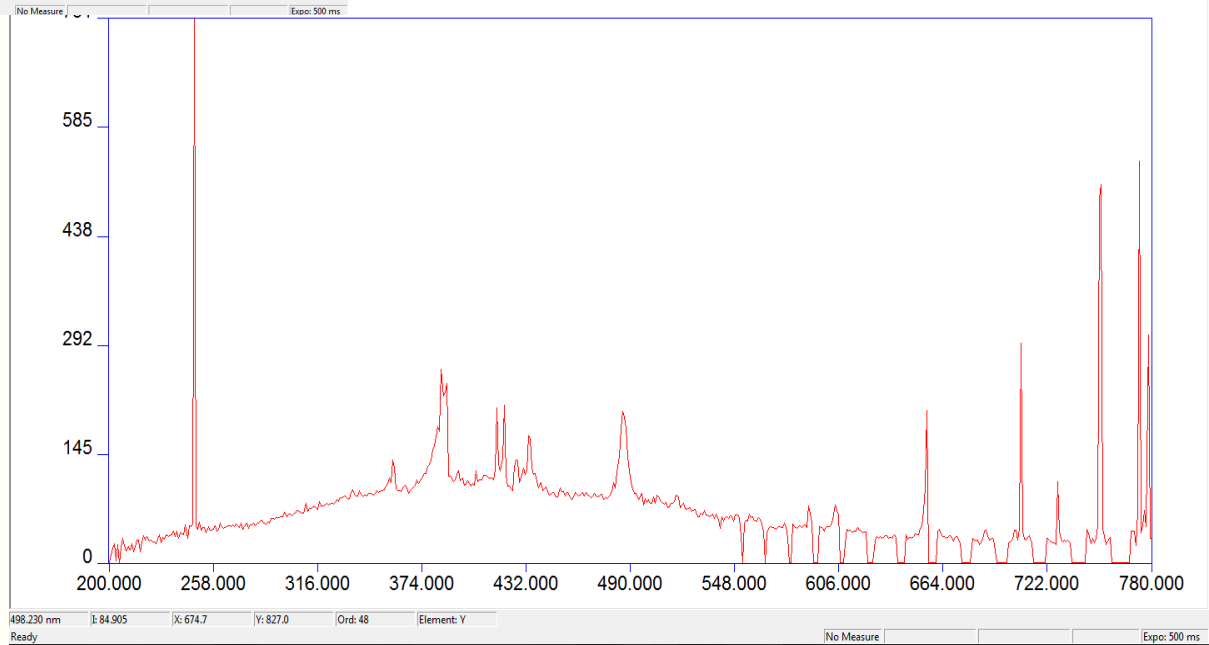


Filter



Window Help

ns | Amplific. 2500 | Apply

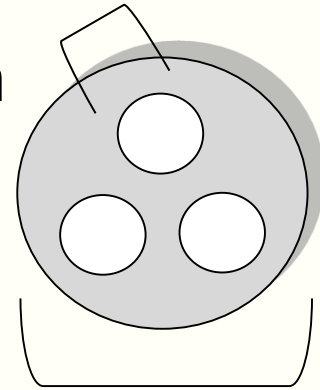


Can bacteria be deposited in a controlled manner?

A steel disk was designed in order to create a reproducible area for bacteria to be placed on.

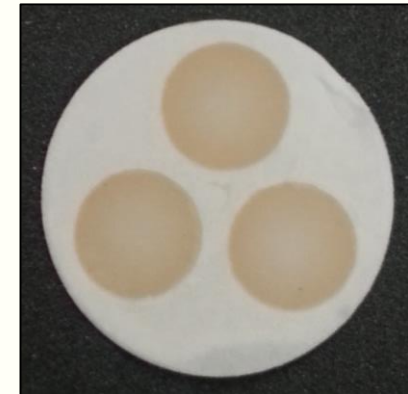


4.3
mm

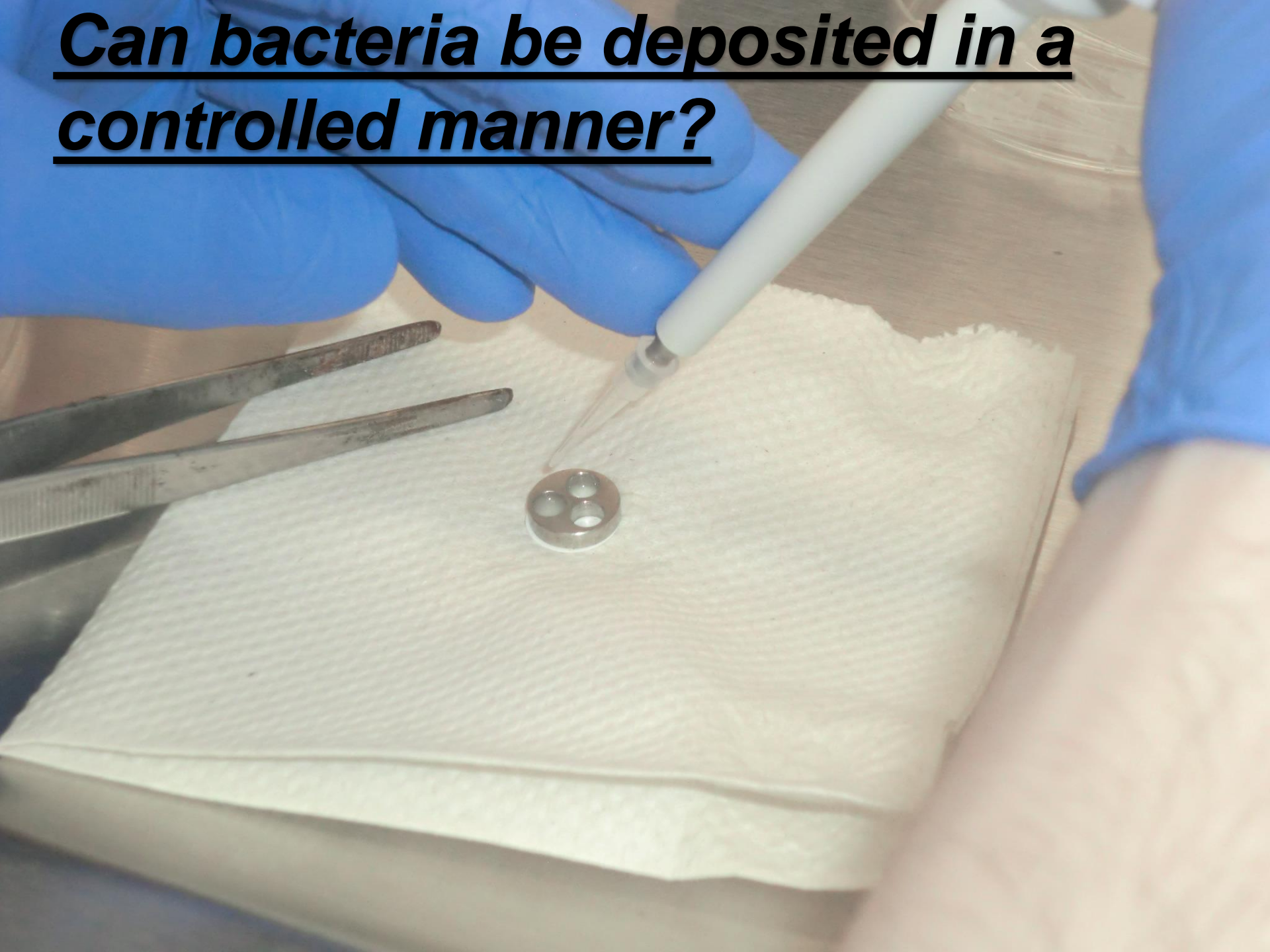


13
mm

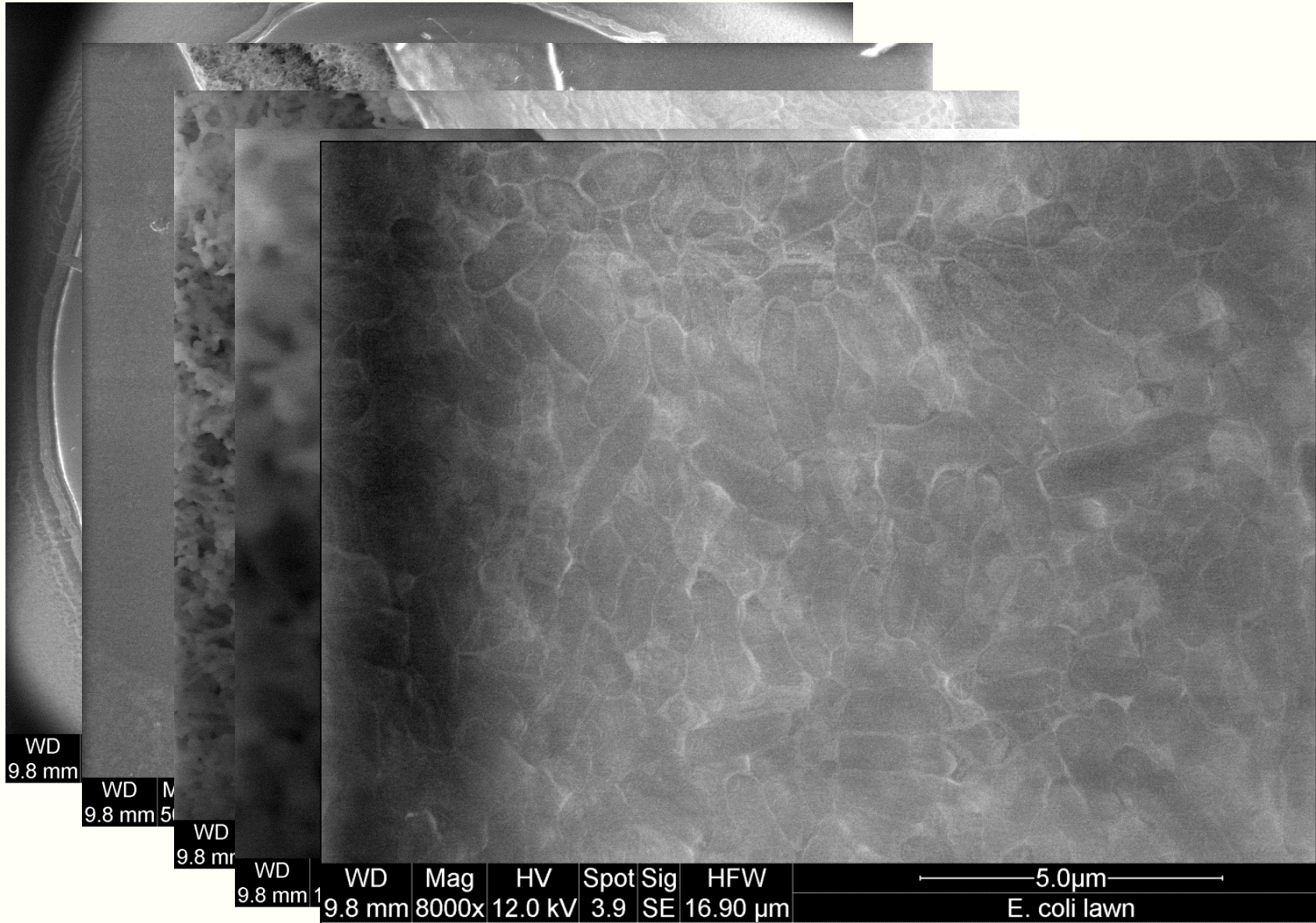
Scanning electron microscopy images verified complete coverage of the deposited bacterial lawn.



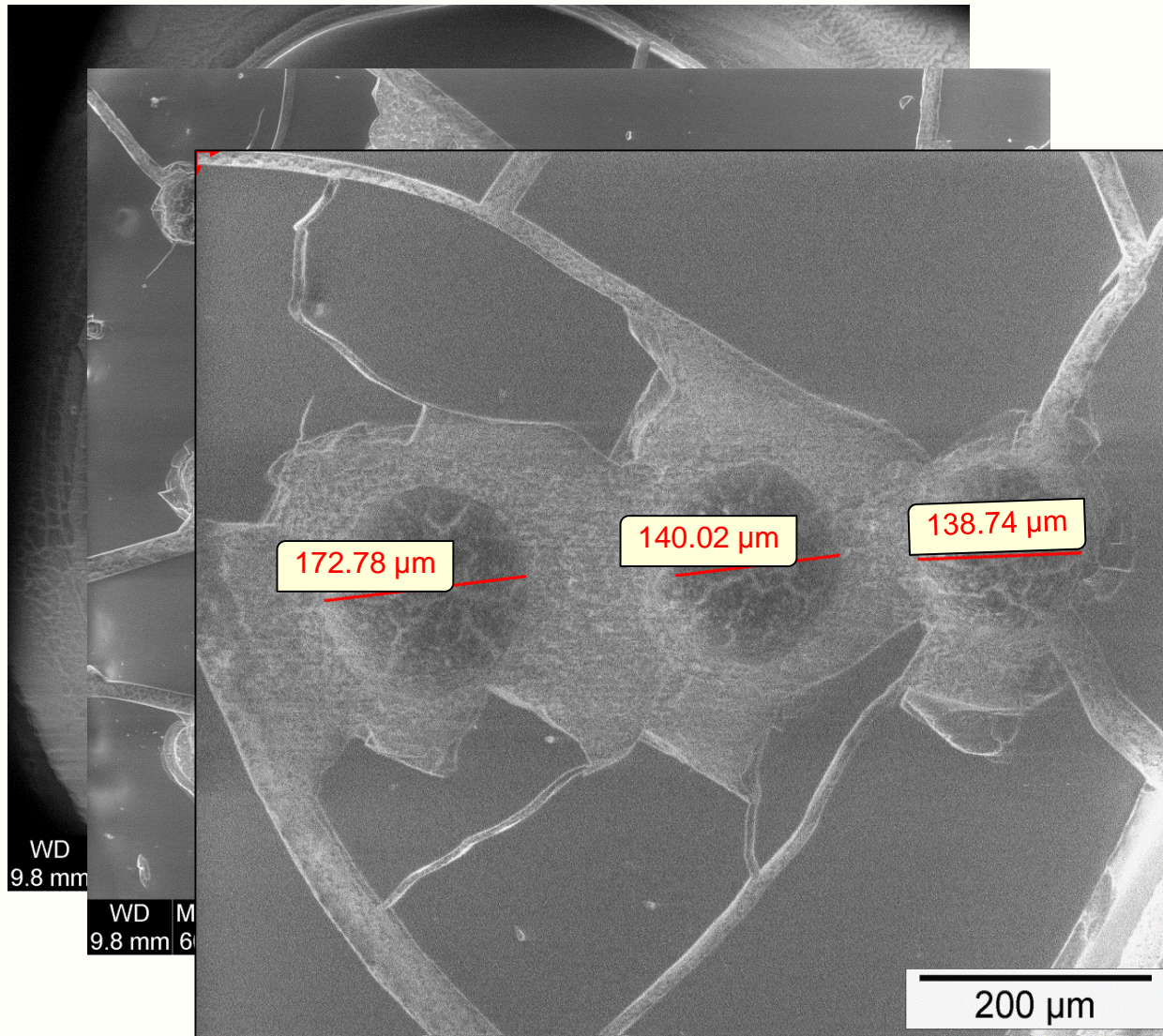
Can bacteria be deposited in a controlled manner?



Can bacteria be deposited in a controlled manner?

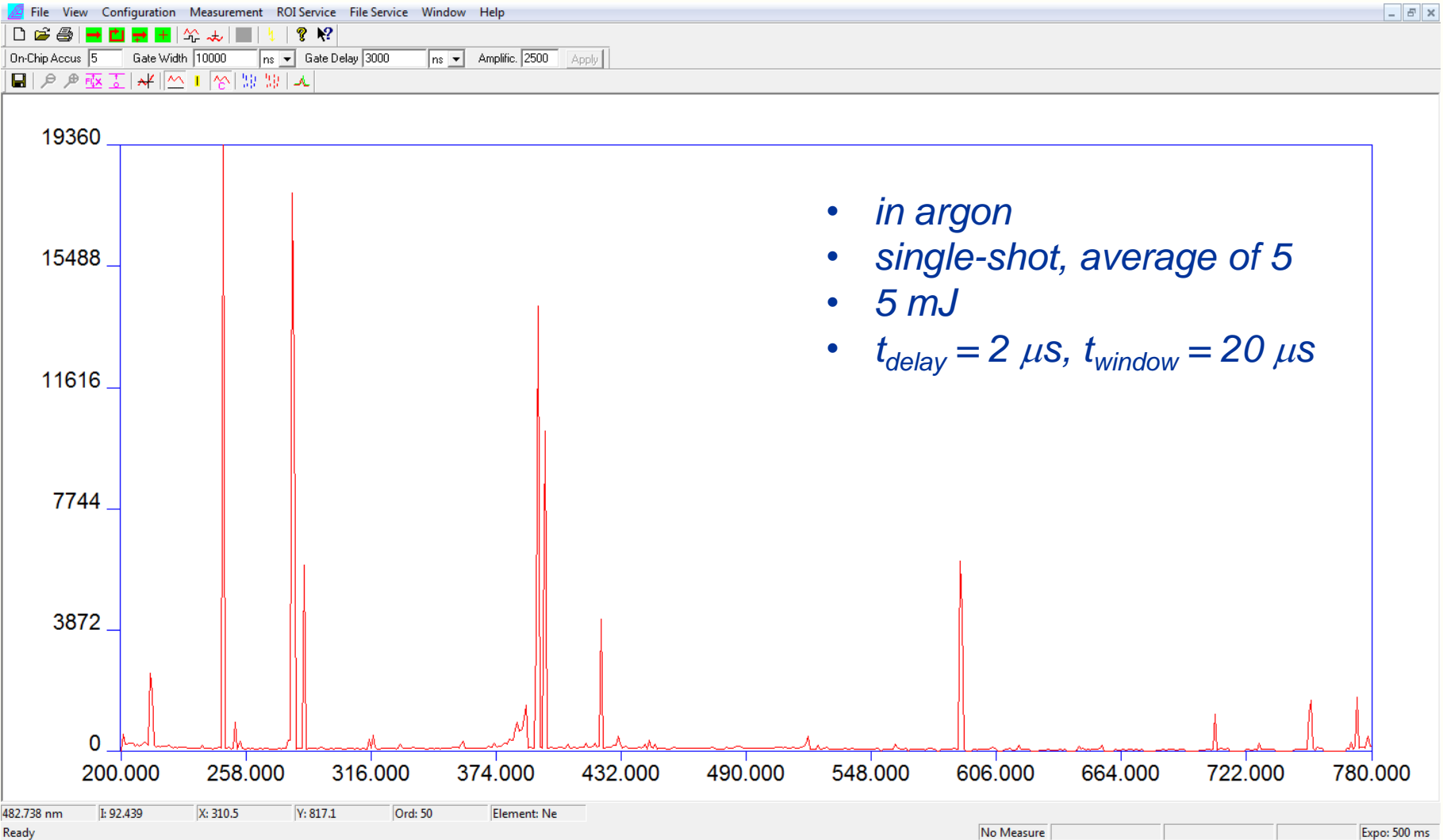


Can bacteria be ablated in a controlled manner?

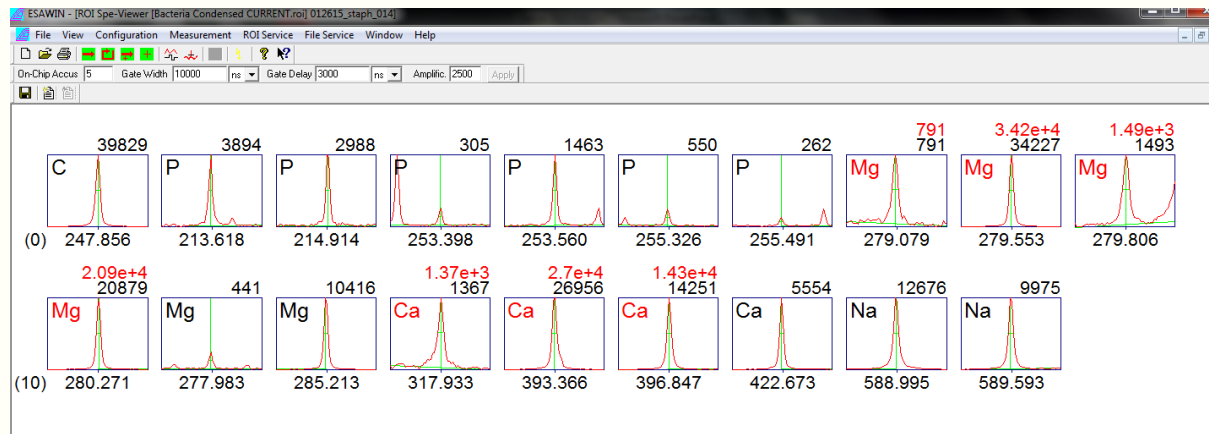


With crater diameters of about 150 μm , the quantity of cells vaporized for each spectrum was estimated at 10^6 (verified using optical densitometry).

Typical Bacterial Spectrum

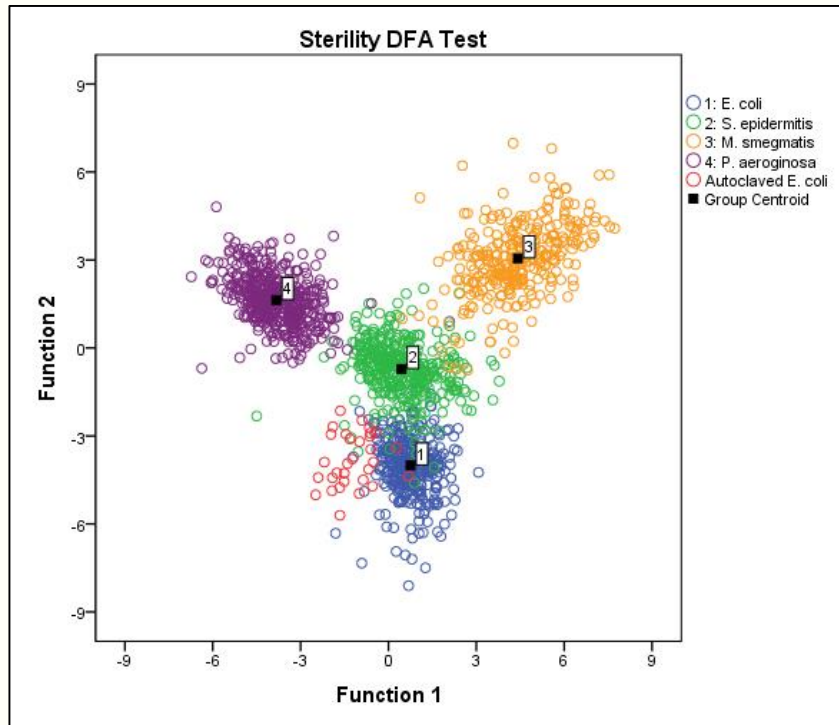


Variable Down-Selection



- New classification model
- 164 independent variable
 - 19 line intensities (all divided by sum)
 - 145 ratios of intensities
- No other metals
 - Farooq (2014) sees S, Cl, Mn, Fe, Al, Cu, etc.
 - Sivakumar (2015) only sees Ca, Na, Mg, K, O, H, C, P
 - We can see other metals when appreciable number of files averaged (to eliminate noise).

Performance With New Library



DFA Classification Grouped by Species

<i>Escherichia</i>	TRUE	FALSE	<i>Staphylococcus</i>	TRUE	FALSE
Positive	98.28%	0.77%	Positive	97.75%	1.44%
Negative	99.23%	1.72%	Negative	98.56%	2.25%
<i>Mycobacterium</i>	TRUE	FALSE	<i>Pseudomonas</i>	TRUE	FALSE
Positive	95.36%	0.33%	Positive	99.57%	0.22%
Negative	99.67%	4.64%	Negative	99.78%	0.43%

Sensitivity: $98 \pm 2\%$ Specificity: $99 \pm 1\%$

PLS-DA Classification Grouped by Species

<i>Escherichia</i>	TRUE	FALSE	<i>Staphylococcus</i>	TRUE	FALSE
Positive	96.55%	1.12%	Positive	96.75%	1.53%
Negative	98.88%	3.45%	Negative	98.47%	3.25%
<i>Mycobacterium</i>	TRUE	FALSE	<i>Pseudomonas</i>	TRUE	FALSE
Positive	97.02%	0.41%	Positive	98.92%	0.33%
Negative	99.59%	2.98%	Negative	99.67%	1.08%

Sensitivity: $97 \pm 3\%$ Specificity: $99 \pm 2\%$

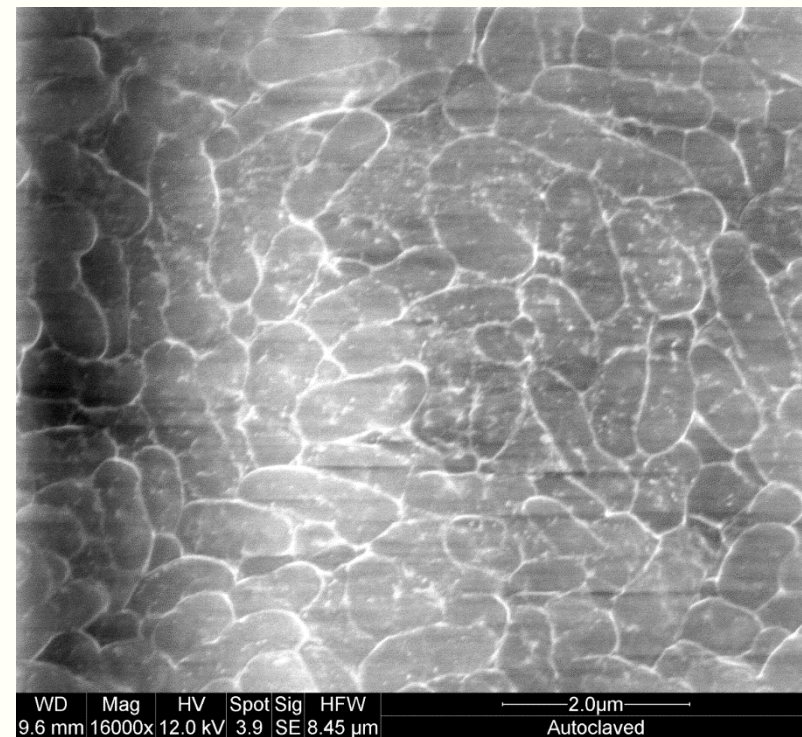
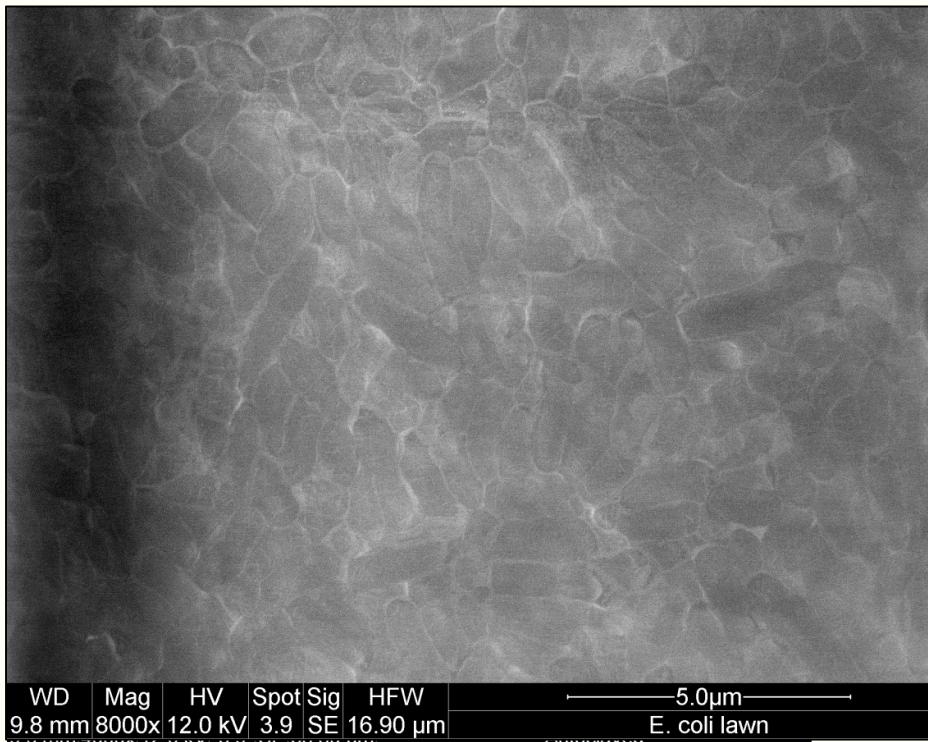
All external validation results

	DFA (by filter)	DFA (by species) above	PLSDA (by species) above
Sensitivity	0.93 ± 0.07	0.98 ± 0.02	0.97 ± 0.03
Specificity	0.98 ± 0.03	0.99 ± 0.01	0.99 ± 0.02

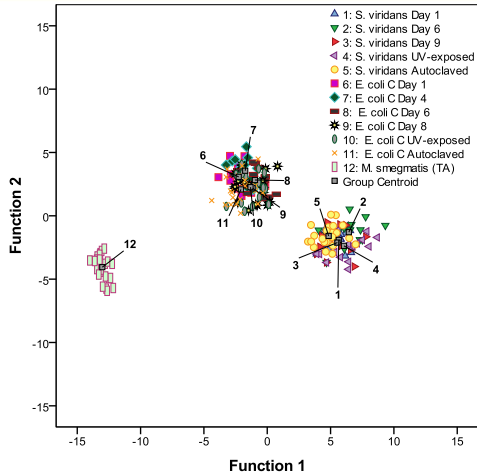
“by filter” means approximately 30 groups in DFA, no relationships between groups assumed

Viability study - Autoclaved

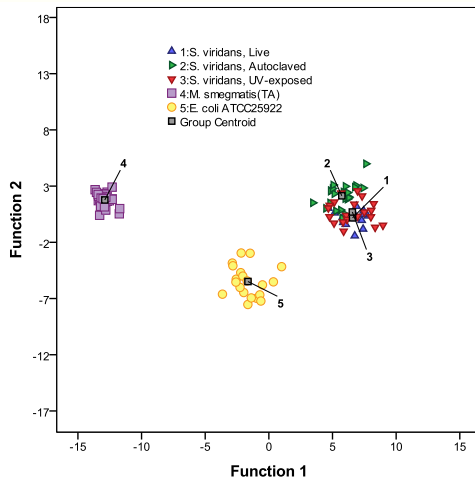
- Live cells were placed in 1 mL of distilled water (microcentrifuge tube).
- This was covered and placed in the autoclave on the liquid cycle.
- Deposited on the filter in the same procedure after vortexing (obtaining solution close to the bottom of the tube).
- Viability was confirmed by restreaking and growth for 24 hours.



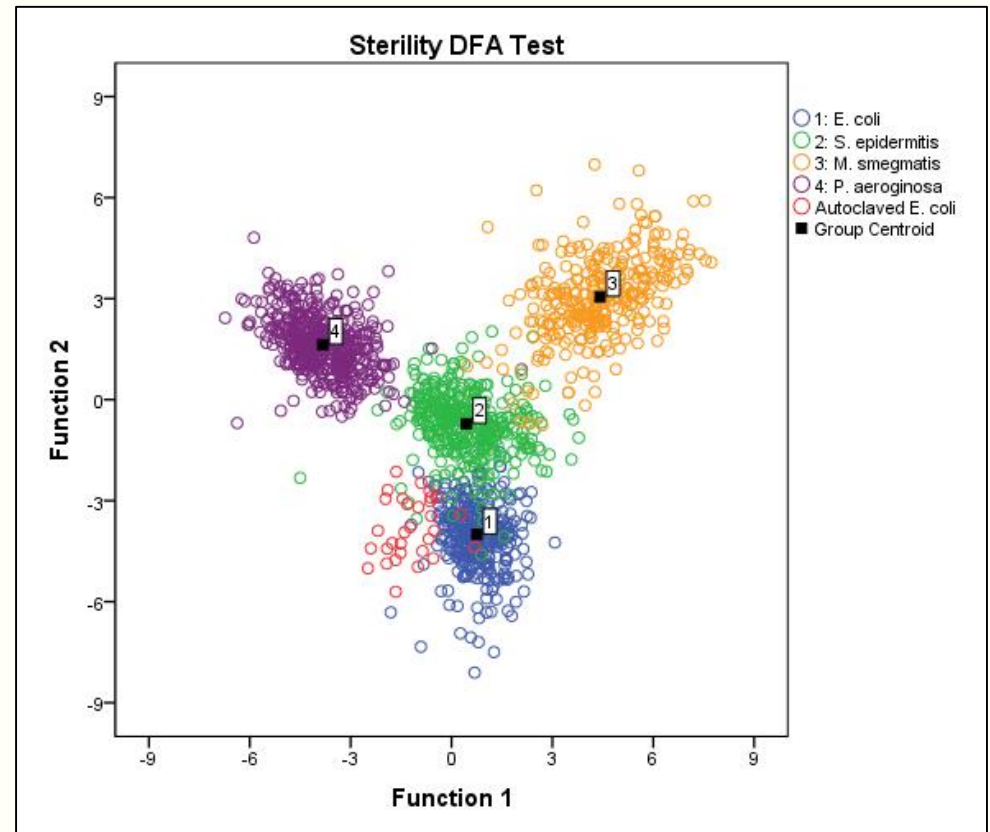
Viability study - Autoclaved



Previous result

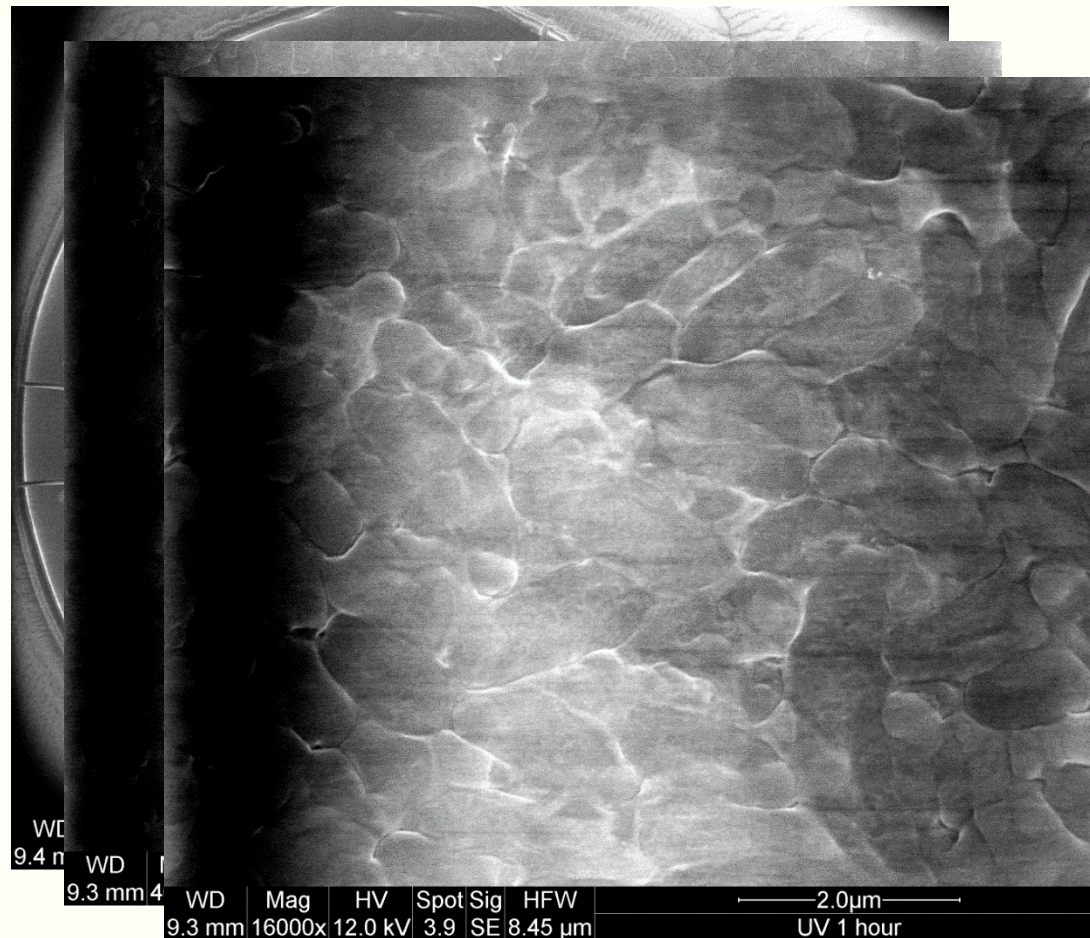


Autoclaved *E. coli* classify >98% as live *E. coli*, yet spectrum seems to be slightly distinct



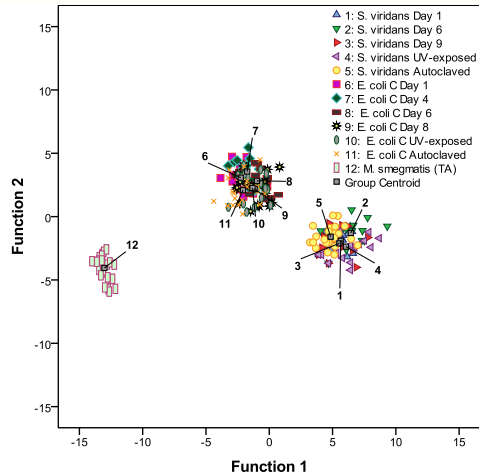
Viability study - UV

- Live cells were deposited on the filter, the whole filter was then placed under UV light (4 W, 254 nm at about 15 cm from the source) for 1 hr, 30 min, 15 min.
- Saw no difference in signal and they classified as live.
- 30 min was chosen for future experiments.
- UV non-viability confirmed via pressing into TSA growth plate and cultured for 24 hours.

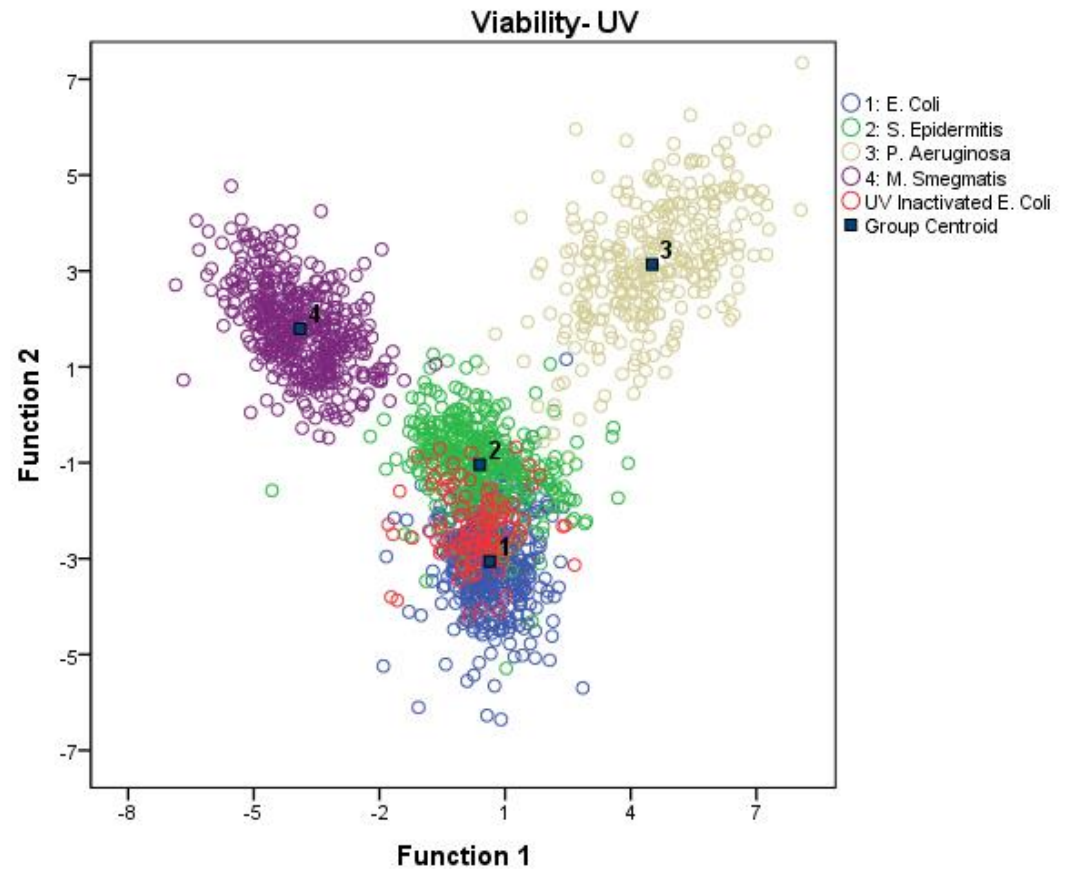
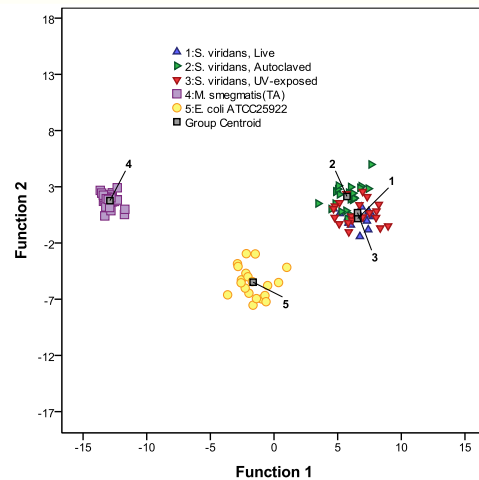


Viability study - UV

Of 118 UV'd *E. coli* spectra, 100% classified as live *E. coli*



Previous result



Summary of Viability Study

Q: So...are live bacteria differentiable from “non-viable” bacteria (autoclaved, UV, sonicated)?

A: Do you want them to be, or don't you?

We see some differences between them (not quantified yet) but they can classify as “live” when comparing against the live species library.

Multari et al. and Sivakumar et al. both DO SEE differences in autoclave (or “heat killed”) and live; and also in sonicated vs. live. Claim is that cell lysis leads to a “leaking out” of material used for discrimination.

Q: If that is true, does our UV result confirm this?

Summary of Viability Study

A: Do you want them to be, or don't you?

Q: Could it be because they are looking for them to be different?

- This leads to more sterilization tests (ongoing), asking the questions:
 - Why are they different (from live)?
 - If they are different from live, are they like each other?
 - If they are like each other, why? Should they be? (structure dependence, nuclear loss, deposition on filter – do they lay the same? Clumping?)

Q: Are the results consistent for all species? Results are just shown for *E. coli* but tests are in progress.

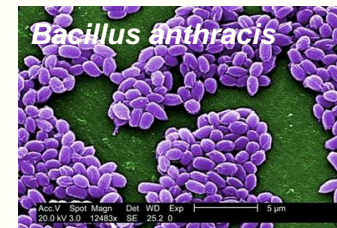
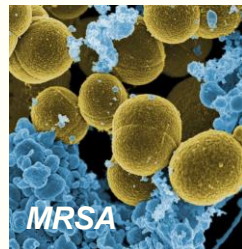
Summary of Viability Study

Q: Lastly, is everyone taking exceptional care to ensure it is not the deposition of the bacteria on their substrates leading to perceived “differences?”

- Bacteria are definitely NOT inorganic microparticles that can be uniformly deposited easily, yielding background-independent LIBS spectra.
- Great care must be taken.

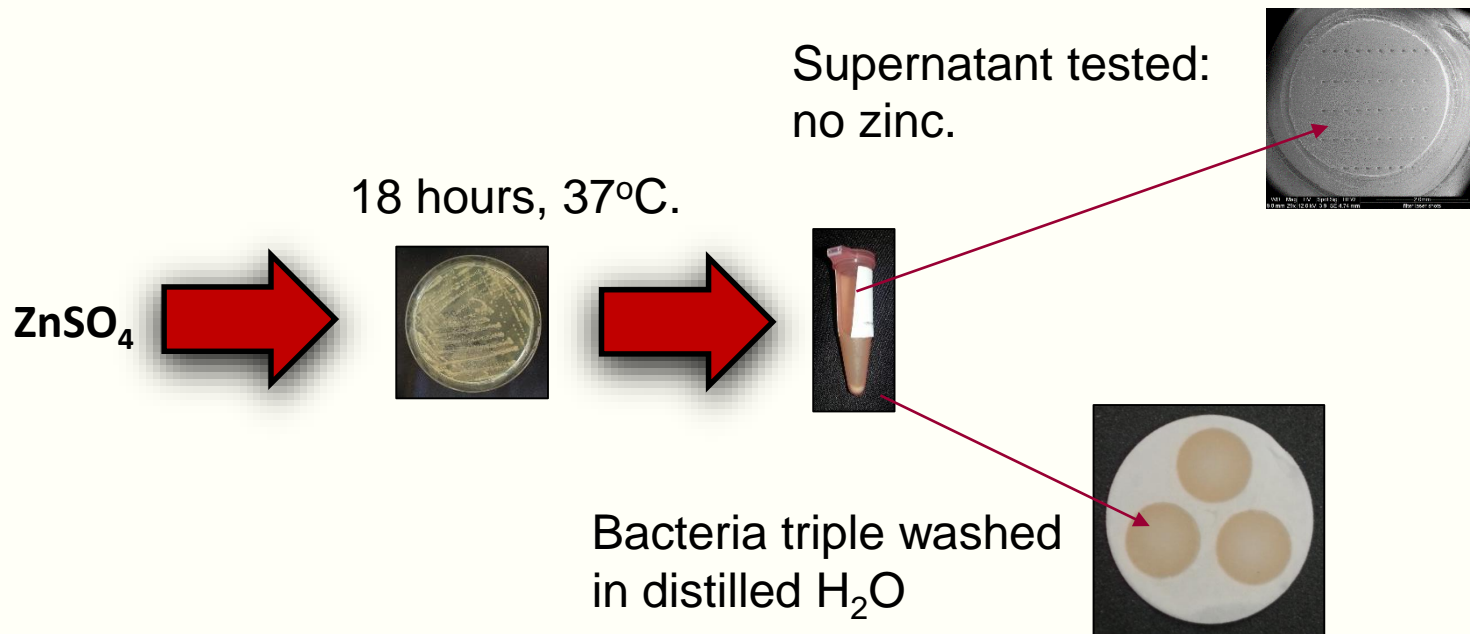


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Intentional Doping of Cells: Zinc

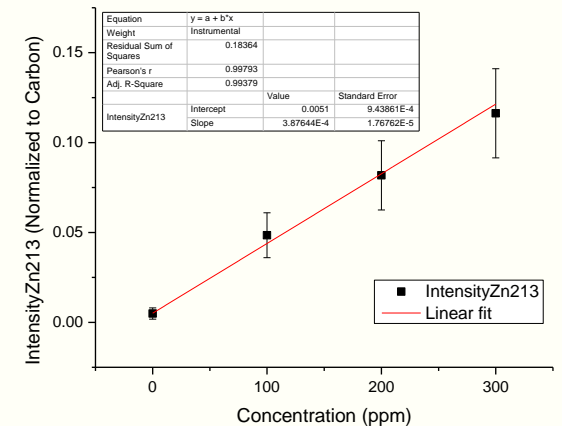
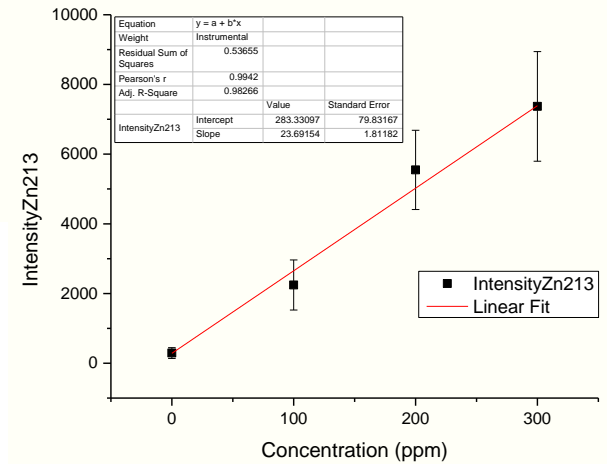
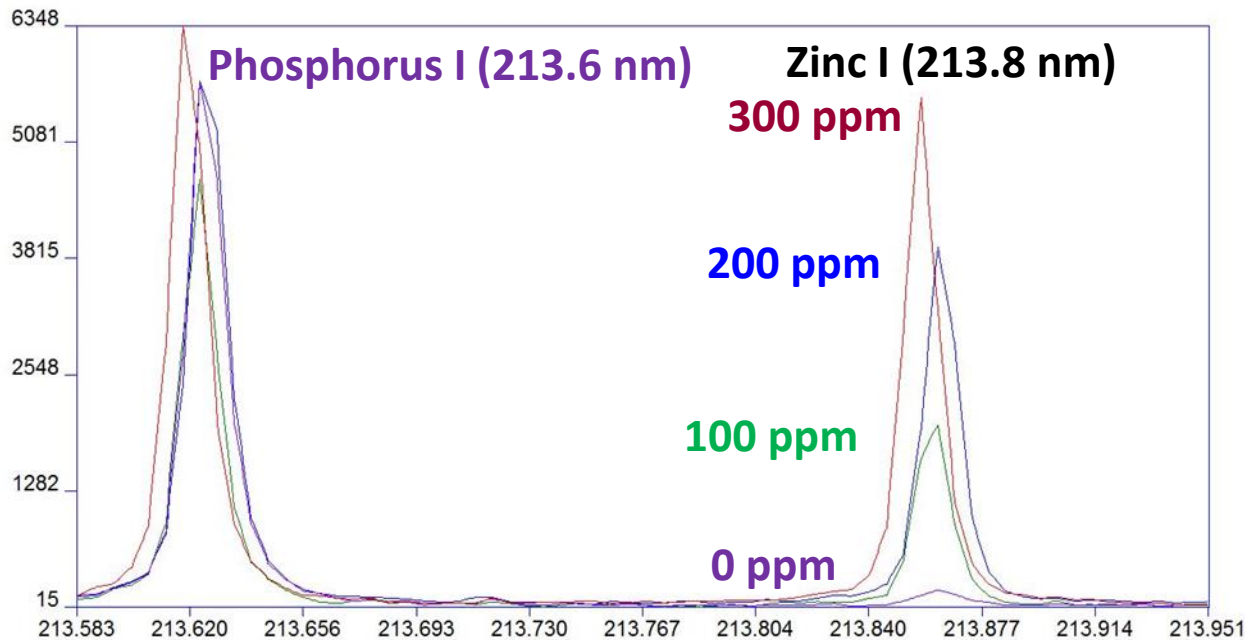
Q: Can metallic elements in the growth environment alter the LIBS spectrum?



Intentional Doping of Cells: Zinc

Q: Can metallic elements in the growth environment alter the LIBS spectrum?

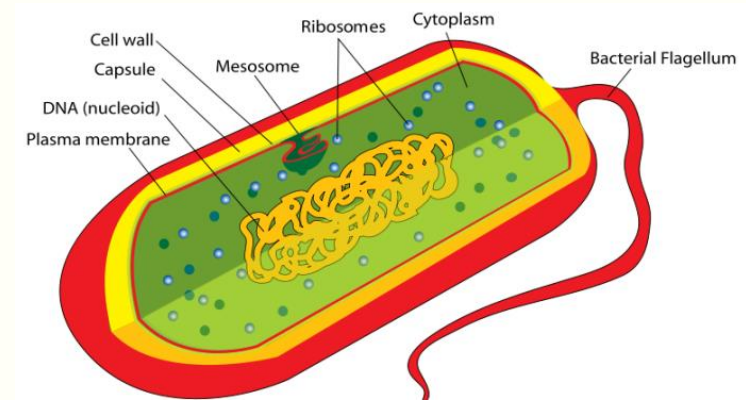
Zinc-doped *E. coli* LIBS spectrum



Intentional Doping of Cells: Zinc

Q: Can metallic elements in the growth environment alter the LIBS spectrum?

A: Yes.

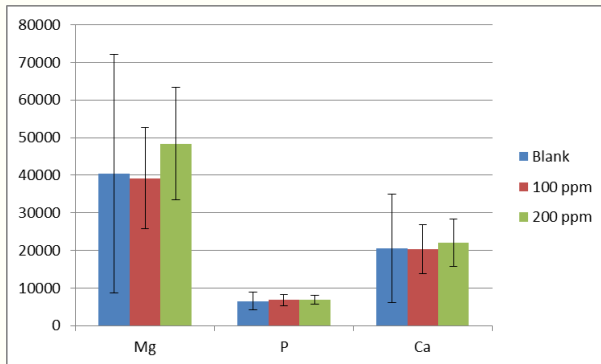


- After rinsing, the zinc signal looked diminished. Is the zinc present inside the cell (in the cytoplasm), or is it merely adsorbed onto the surface?
- Magnesium lines seemed diminished. Zinc and magnesium are comparable in size, and are both divalent cations, and the cell possibly substitutes Zn for Mg when Zn is present in abundance in the growth medium.

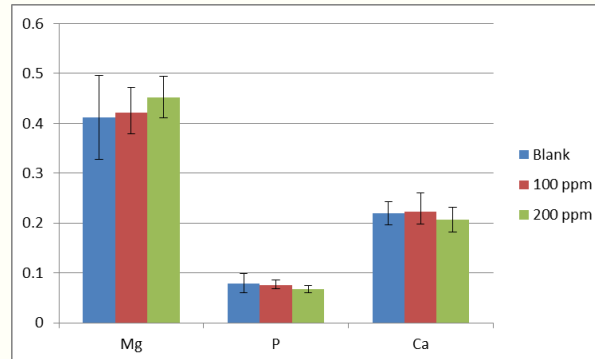
Intentional Doping of Cells: Mg

To test whether significant changes in magnesium concentrations could be achieved within limits of extremely hard water (high magnesium concentrations), we carried out the following experiment:

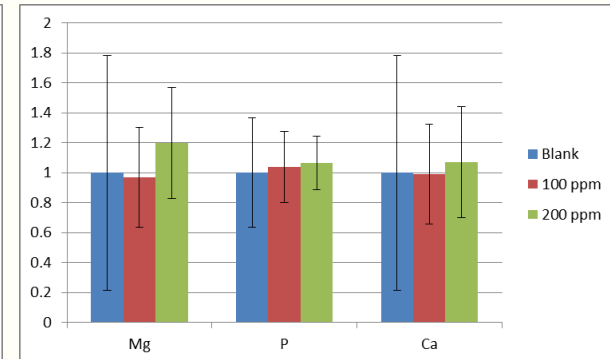
Label	Concentration of Magnesium (ppm)	Magnesium Lines in Spectra
B	0	Normal Mg lines
S1	<0	No bacterial growth
S2	100	Very slightly elevated peaks
S3	200	Very slightly elevated peaks



Raw LIBS intensities



As fraction of total spectrum



Normalized to blank

- 200 ppm cultures classified as control *E. coli*.
- Physiologic concentration is 18-30 ppm.

Impact of blood glucose/metal uptake

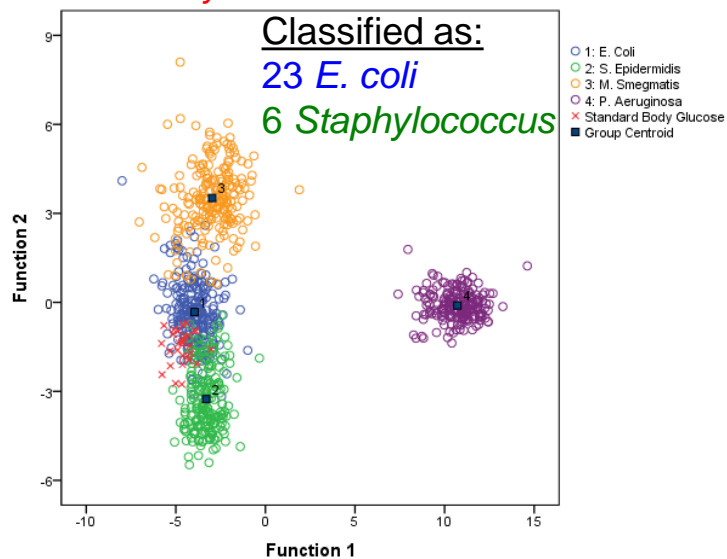
- Bacterial uptake of metal ions is a function of their metabolism, since most metals need active transport to be imbibed.
- Bacteria that grow faster/more rapidly should theoretically show higher metal levels than slower ones (of the same species).
- Perhaps the slight difference in osmotic pressure caused by high sugar content in the environment could cause a change in metal ion concentrations in the bacterial cells?
- In diabetics, blood sugar levels fluctuate rapidly, ranging from very high to very low within a few hours of eating.

We designed an experiment to verify the robustness of DFA classification of bacteria grown in various blood sugar concentrations by doping the growth medium with excess glucose and checking the change in metal ion signals in the LIBS spectra.

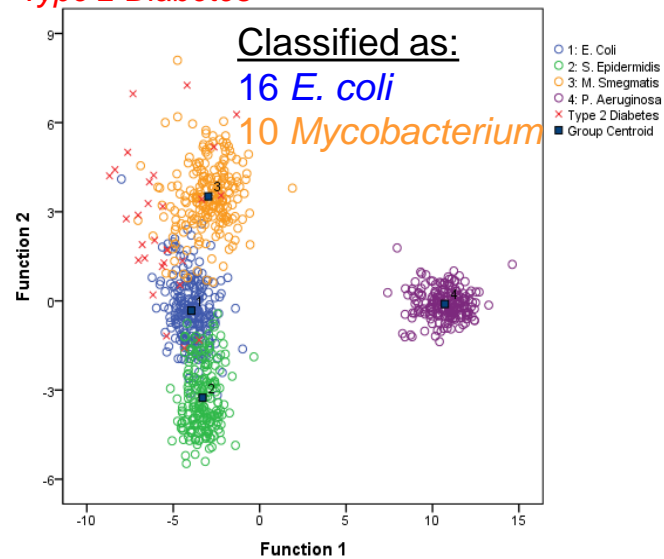
Label	Concentration of Glucose (g/L)	
B	1.35	Control
S1	1.44	Type II diabetes
S2	1.62	Type I diabetes
S3	1.8	Children's type I diabetes

Impact of blood glucose/metal uptake

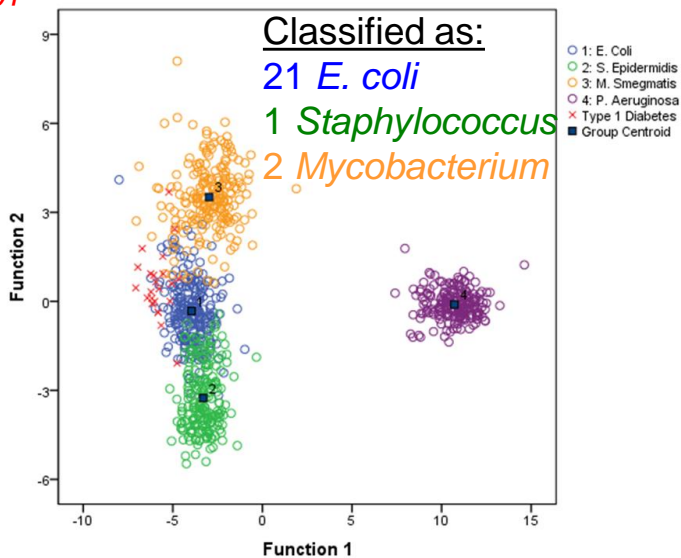
Standard Body Glucose



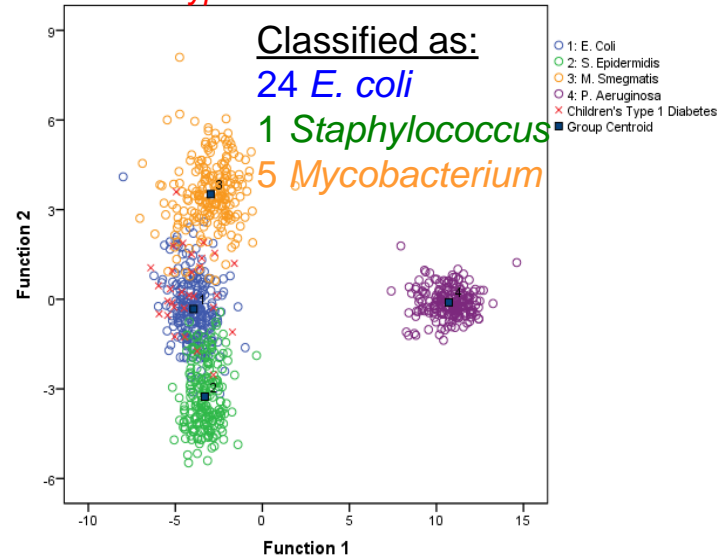
Type 2 Diabetes



Type 1 Diabetes



Children's Type 1 Diabetes



So many questions...

...but all tests to date have proven the possibility of using LIBS for a rapid pathogen diagnostic, as well as numerous other biomedical applications.

Work continues, with generous help from:

- University of Windsor



- NSERC Discovery Grant



Natural Sciences and Engineering
Research Council of Canada

Conseil de recherches en sciences
naturelles et en génie du Canada

- CFI-LOF grant

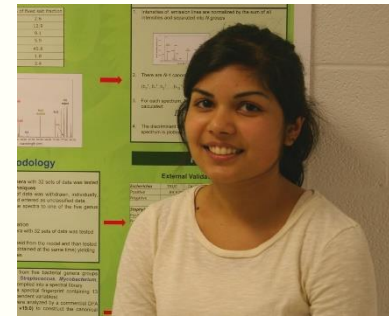
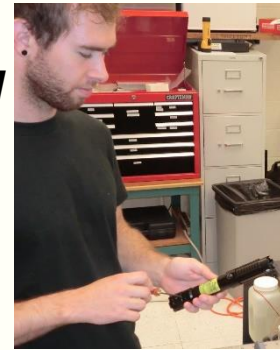


All Credit to the Students



Dan Trojand

Russell Putnam



Khadijia Sheikh



Siddharth Doshi

Anthony Piazza

Dylan Malenfant

Derek Gillies

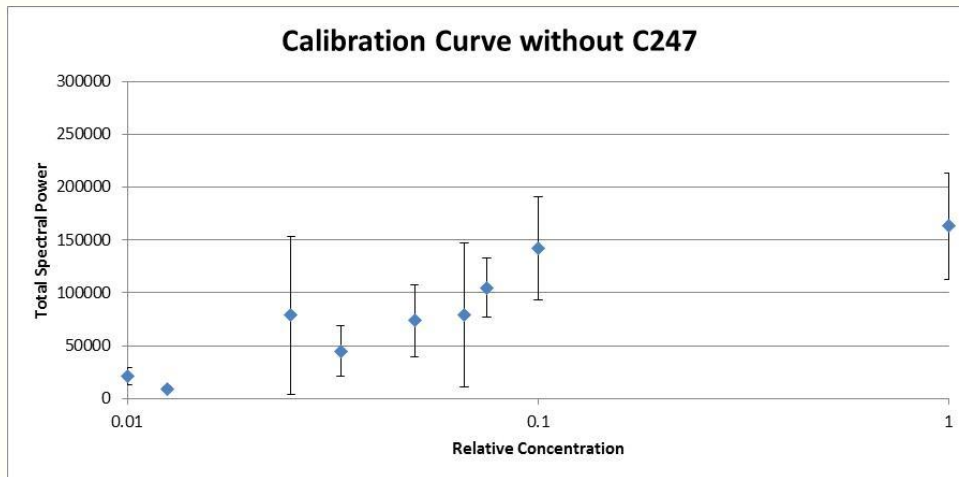
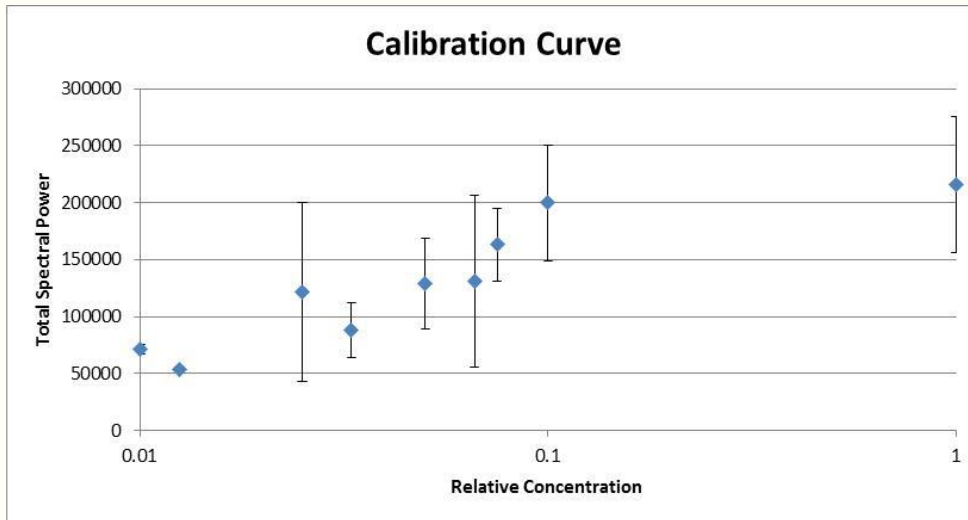
Allie Paulick



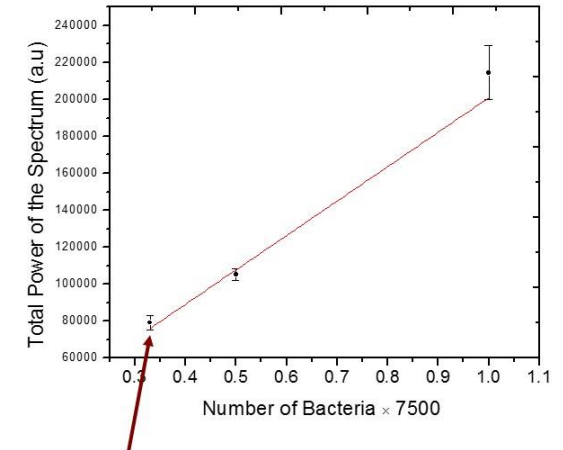
Andrew Daabous



New Concentration Study



Previous result



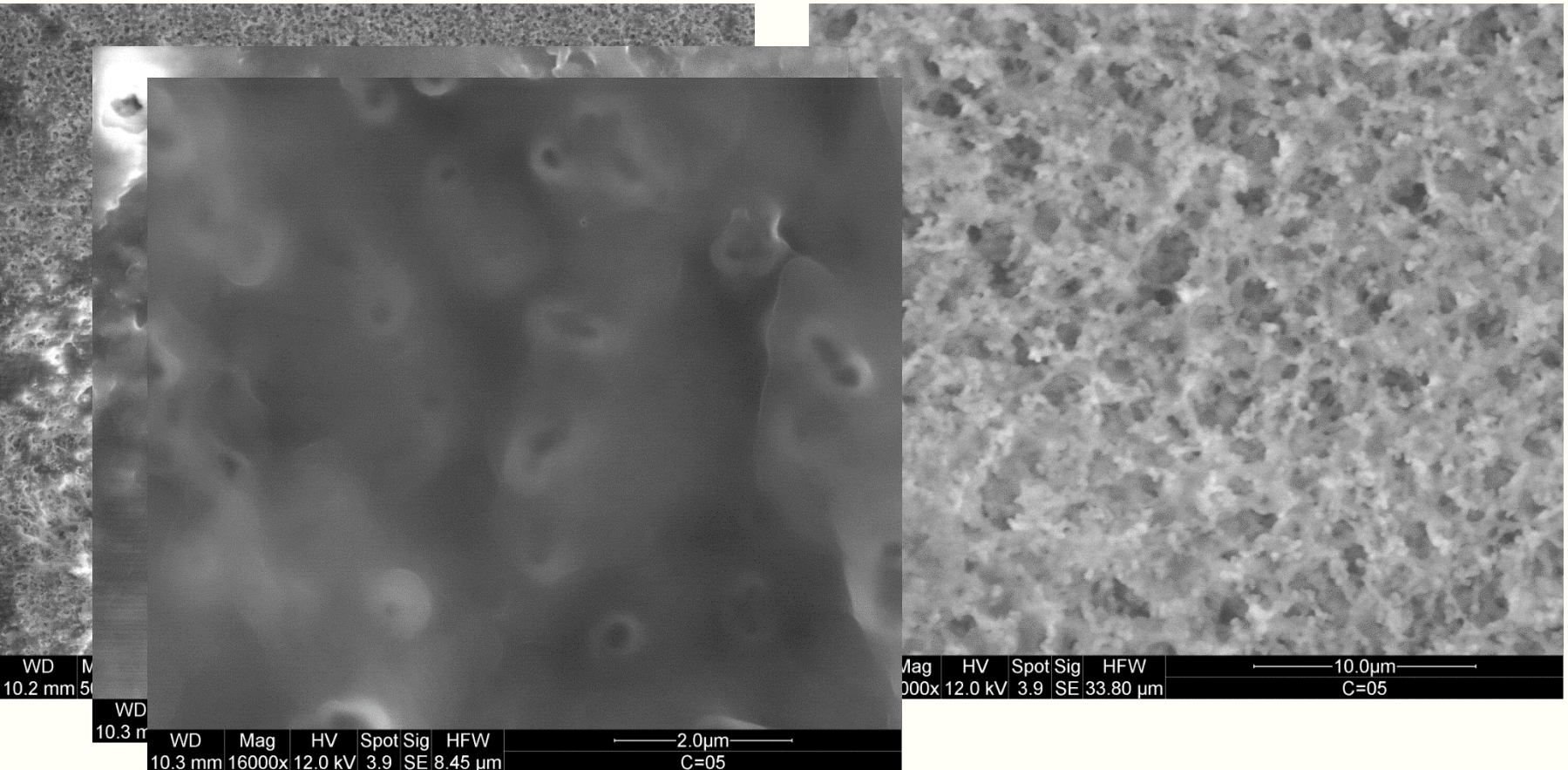
5 laser sampling locations

~500 bacteria per locations

- Performed with serial dilutions.
- “Concentration 1” → harvest entire plate of colonies off TSA, suspend in 1.5 mL distilled H₂O
- Measure with optical densitometry
- OD=0.1 measured for C=0.001 (from literature OD 0.1=10⁸ cells/mL).
- C=1 → 10¹¹ cells/mL
- Implies for C=1, 10⁶/shot

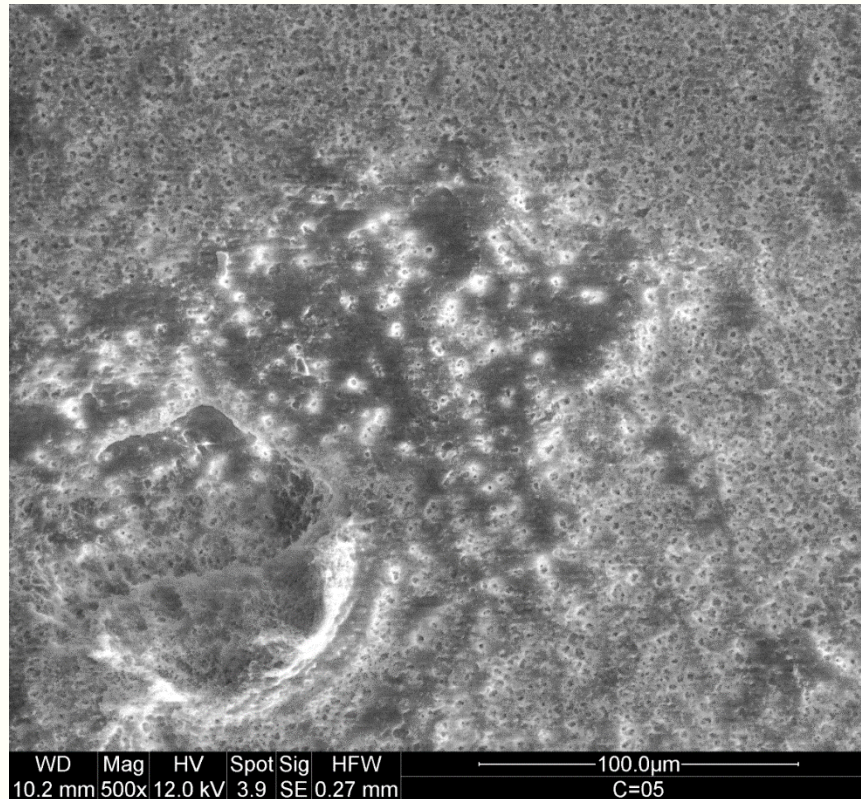
Viability study - Sonication

- Cells were placed in 1 mL distilled water (microcentrifuge tube) sonicated at a setting of 3 W for 10 seconds.
- Deposited on the filter in the same procedure after vortex mixing (obtaining solution close to the bottom of the tube).
- When done the second time, 0.8 W for 10 5s-pulses



Viability study - Sonication

- Sonicated *E. coli* tested at multiple concentration (c=0.05, =0.025). Both classified as autoclaved *E. coli* over live.
- Sonicated cells were clearly disrupted and rather than assisting in sample homogeneity, showed increased clumping/heterogeneity



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Society for
Applied Spectroscopy

North American
Society for Laser-Induced
Breakdown Spectroscopy

SciX - Right Size, Right Science, Right Conference

Member Organizations

American Chemical Society's Analytical Division

AES Electrophoresis Society

American Society for Mass Spectrometry

ANACHEM

Coblentz Society

Council for Near Infrared Spectroscopy

Infrared and Raman Discussion Group (IRDG)

International Society of Automation - Analysis Division

Royal Society of Chemistry Analytical Division

The North American Society for Laser-Induced

Breakdown Spectroscopy

Society for Applied Spectroscopy

The Spectroscopical Society of Japan (SPSJ)

TECHNICAL PROGRAM – MONDAY
Orals 1:20 – 3:00 pm and 3:50 – 5:30 pm

Monday Afternoon, Room 551A
NUCLEAR FORENSICS
Organizer and President: Andrew Duffin

- 3:50 (181) **Applications of a New Single Stage Accelerator Mass Spectrometer to Trace Detection and Nuclear Forensics**; Albert Fahey¹, Kamron Fazel¹, Kenneth Grabowski¹, Evan Groopman¹; Naval Research Laboratory
- 4:10 (182) **X-Ray Microscopy of Nuclear Materials**; Jesse Ward¹, Greg Eiden¹, Andrew Duffin¹; ¹Pacific Northwest National Laboratory
- 4:30 (183) **Advances in Analysis of Samples for Nuclear Non-Proliferation at CEA/DIF**; Bruno Bernard-Michel¹, Fabien Poincurier¹, Maxime Bridoux¹, Anne-Laure Faure¹, Amélie Hubert¹, Olivier Marie¹, Anne-Claire Pottin¹; ¹CEA-DIF, Bruyères le Châtel
- 4:50 (184) **Discrimination of Uranium ore Concentrates from Several Countries by Chemometric Data Analysis**; Josette El Haddad¹, Aissa Harhira¹, Alain Blouin¹, Mohamed Sabsabi¹, Marvin Zalusk², Chunsheng Yang³, Christopher Drummond⁴, Slobodan Jovanovic⁵, Tanya Hinton⁶, Ali El-Jaby⁷; ¹National Research Council Canada - Energy, Mining and Environment; ²National Research Council Canada - Information and Communications Technologies; ³Canadian Nuclear Safety Commission
- 5:10 (185) **DC Arc Spectroscopy – Plasma Characterization for Direct Solid Analysis of Nuclear Materials**; Benjamin T. Manard¹, John Matonic¹, Robert Jump¹, Dennis Montoya¹, Alonso Castro¹, Ning Xu¹; Los Alamos National Laboratory

Monday Afternoon, Room 556B
NEAR IR
Organizer and President: Franklin E. (Woody) Barton

- 3:50 (186) **NIR With Problem Data Sets**; Franklin Barton¹, James de Haseth¹; ¹Light Light Solutions Instruments, Inc.
- 4:10 (187) **A New Look at the Derivative Quotient Method in Regression**; David Hopkins¹, Karl Norris²; ¹NIR Consultant, Battle Creek, MI; ²NIR Consultant, Beltsville, MD
- 4:30 (188) **Ultra-Compact Smart Spectrometers For Food, Agriculture, and Pharmaceutical Applications**; Nada O'Brien¹, Christopher Pederson¹, Peng Zou¹; ¹JDSU Corporation
- 4:50 (189) **A Novel Configuration for Near-Infrared Analysis of LPG Composition and Quality Control in a Refinery Setting**; Susan Fouk¹, Shashi Mistry², Terry Todd¹, Nate Peters¹, Dian Wang³; ¹Guided Wave, Inc.; ²Suncor Energy
- 5:10 (190) **Field Analysis of Fuel using a Portable Near-Infrared Spectrometer**; Wayne Smith¹, Carl Brouillette¹, Chetan Shende¹, Stuart Farquharson¹; ¹Real-Time Analyzers, Inc.

Monday Afternoon, Ballroom E
LIBS FOR PHARMACEUTICAL AND BIOLOGICAL DIAGNOSTICS
Organizer and President: Lydia Breckenridge

- 3:50 (191) **Recent Progress and Current Challenges in Using LIBS for Bacteriological Identification**; Steven Reese¹, Dylan Malenfant¹, Derek Gillies¹, Viora Riberdy¹, Anthony Piazzas¹; ¹University of Windsor
- 4:30 (192) **Laser-Induced Breakdown Spectroscopy for the Evaluation of Residual Catalysts in Pharmaceuticals**; Lydia Breckenridge¹; ¹Bristol-Myers Squibb

- 4:50 (193) **Identification of Meat Species by using Laser Induced Breakdown Spectroscopy**; Gonca Bilge¹, Banu Sezer¹, Hasan Murat Velioglu², Kemal Efe Eseller³, Halil Berberoğlu⁴, İsmail Hakkı Boyacı¹; ¹Hacettepe University, Department of Food Engineering; ²Namık Kemal University, Department of Agricultural Biotechnology; ³Atılım University, Department of Electrical & Electronics Engineering; ⁴Gazi University, Department of Physics
- 5:10 (194) **Study of Plasma and Identification of Hazardous Elements in the Polystyrene using Laser Induced Breakdown Spectroscopy**; W. Aslam Farooq¹; ¹King Saud University

Monday Afternoon, Room 552B
PHARMACEUTICAL APPLICATIONS OF LOW WAVENUMBER SPECTROSCOPY
Organizer and President: James Carriere

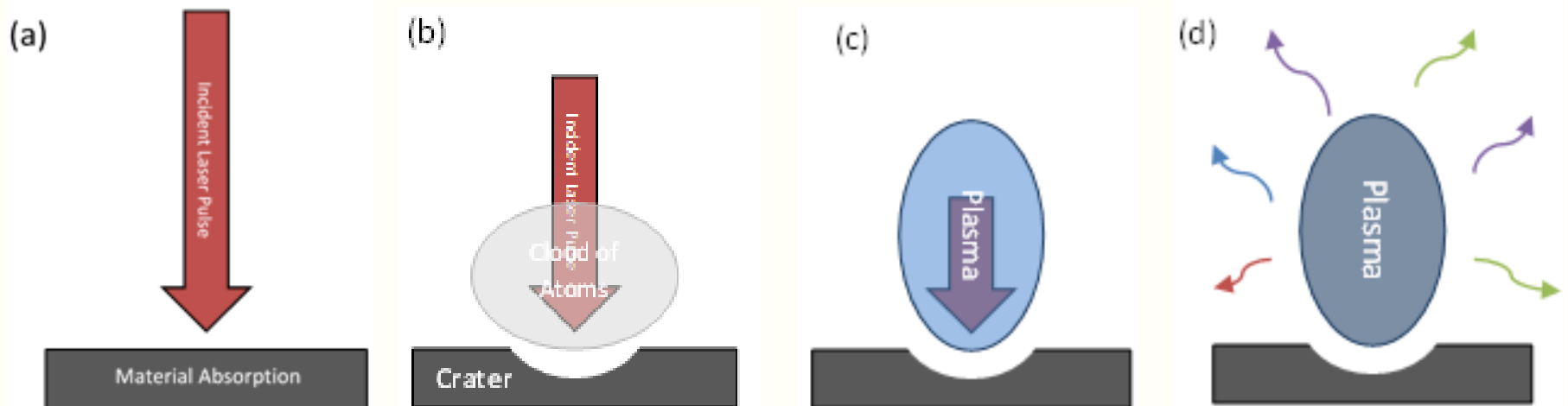
- 3:50 (195) **Application of Low Frequency Raman During the Crystallization Process**; John Wasylivk¹, Ming Huang¹, Robert Wethman¹; ¹Bristol-Myers Squibb Co.
- 4:10 (196) **The Contribution of the Low-Frequency Raman Spectroscopy to the Structural Description of Disordered Molecular Systems and Their Transformations: Application to Pharmaceuticals**; Alain Hedoux¹, Laurent Paccou¹, Yannick Guinet¹; ¹University Lille 1, UMET - UMR CNRS 8207
- 4:30 (197) **Chemical Imaging of Crystalline Components in Pharmaceutical Dosage Forms by Using Low Frequency Raman Spectroscopy**; Toshiro Fukami¹, Motoki Inoue¹, Hiroshi Hisada¹, Tatsuo Koide²; ¹Meiji Pharmaceutical University; ²National Institute of Health Sciences
- 4:50 (198) **Calibration of a Terahertz Analyzer for Predicting Solid Fraction in Roller-Compacted Ribbons and Tablets in a Small-Scale Piloting Study to Facilitate Pharmaceutical Formulation Development**; Mark Sullivan¹, Elaine Harrop Stone², Mouwara Hoque², Xiao Hua Zhou¹, Richard McKay¹; ¹Advantest America Inc; ²Merlin Powder Characterisation Ltd
- 5:10 (199) **Low Wavenumber Raman Spectroscopy Applications in API Phase Discovery and Characterization**; Courtney Maguire¹, Andrew Brunskill¹; ¹Merck Research Laboratories

Monday Afternoon, Room 555B
HOT TOPIC DISCUSSION SESSION – TERS RESOLUTION
Organizers: Duncan Graham, Pavel Matousek, and Ian Lewis;
President: Duncan Graham

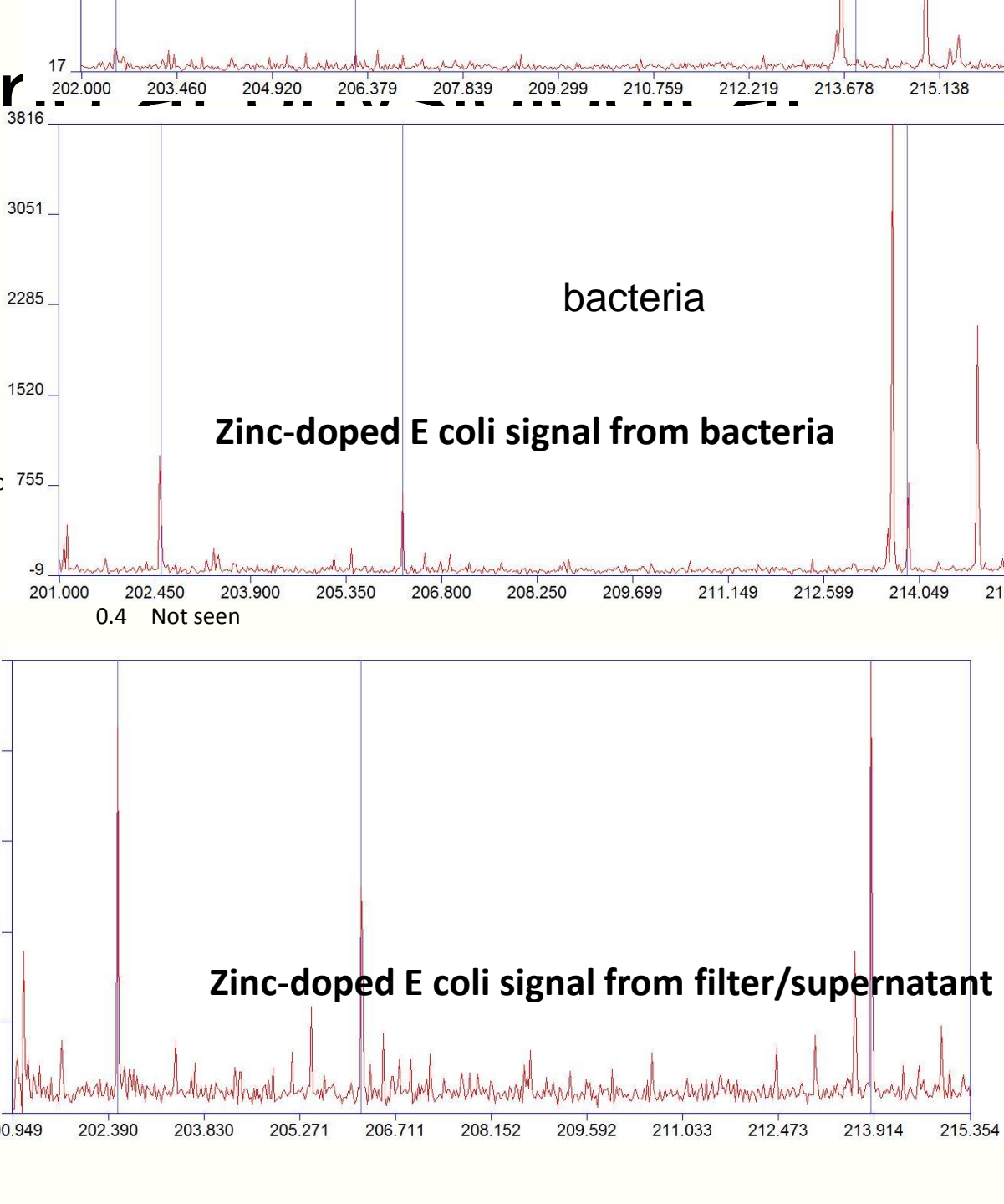
- 3:50 (200) **Recent Advances in Tip-Enhanced Raman Spectroscopy**; Richard Van Duyne¹; ¹Northwestern University
- 4:10 (201) **Resolution and Enhancement in TERS Microscopy**; Satoshi Kawata¹, Atsushi Taguchi¹; ¹Osaka University
- 4:30 (202) **Molecular Structure Changes on the Nanometre Scale Investigated and Induced by TERS**; Volker Deckert^{1,2}; ¹University of Jena; ²Leibniz Institute of Photonic Technology
- 4:50 **Discussion**

What is LIBS?

- Intense laser pulse interacts with the target material and energy is absorbed.
- Energy absorbed results in heating and vaporization of the material. Matter from the surface is removed and a vapor is formed.
- Laser pulse is still incident on the vapor. Energy is absorbed, inducing heating and plasma formation (with temperature of $\sim 50,000$ K).
- As the plasma cools, atoms/ions/molecules decay by spontaneous emission. Light is collected and dispersed by the spectrometer. Resulting spectrum is analyzed to identify elements present in the target material.



Zinc dopir



Label
B
S1
S2
S3

Concentration c